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Orekhovich, V. N., Konikova, A. S., Orekhovich, K. D., and Dobbert, N. N. *Concerning the Metabolic Turnover Rates of Various Organ and Tissue Proteins. Akademiia Nauk (SSR Doklady, 71:105, 1950.

Bresler, S. Y., Finogenov, P. A., and Frenkel, S. Y. A discussion of the

structure of the Magromolecule of Procollagen. Reports of the Academy of Sciences of the U.S.S.R., LXXII (72), 3, pp. 555-8, 19: 12, A. A. "Concerning the Proteins of Skin". Biokhimiia, 12 (4):

Tustanovskir, A. A. "Con 285-290, 1947.

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3. Two other Soviets, S. L. Pupko and A. L. Zaydes have also written on electronmicroscopic investigations of collagen published in proceedings of the Academy of Sciences

> Zaydes, A. L. and Pupko, S. L. Electron-Microscopic Investigation of the Effects of Alkalis and Pancreatin on Collagen". Akademiia Nauk USSR Doklady, Vol. 73: 991, 1950. "The Electron-Microscopic Examination of

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The Electron-Microscopic Examination of Collagen Using the Replica Technique". Akademiia Nauk USSR Doklady, Vol. 73: 379, 1950.

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The fact that they are members of the Central Research Institute of the Leather Footwear Industries leaves little doubt as to the Soviet belief in the future of reconstituted collagen.

- US scientists reviewing this work found the Soviet conclusions particularly interesting. For the past three years 49-52/ a group of US scientists has been working on the ultrastructure of connective tissue based on the work of the above named Soviet scientists. The US group also hopes to gain information on the formation and nature of collagen and determine its molecular structure. To date attempts of US scientists to make contact, through the Soviet Academy of Science, with the Soviet collagen specialists have been unsuccessful. The Soviets also have never mentioned US work in the field; so it is hard to assay Soviet progress.
- In these days of the successful use of such agents as cortisone, many people are getting on the bandwagon to investigate connective tissue, and the field is becoming increasingly more popular. Results of this research will find application in medicine and may give insight into such phenomena as wound healing and aging. The production

25X1	of	synthetic	fiber	and	synthetic	leather	as	а	result	of	this	research	is	quite	possible.
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8. This side of the Iron Curtain, the British are probably the leaders in the field.

Germany and Scandinavia are also very active in tissue ultrastructure. There seems to be some work going on in Italy.

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Bresler, 5.4., P. A. Finogenov, and S.Y. Frenkel. A discussion of the structure of the Macromolecule of procollagen. Reports of the Academy of Sciences of the U.S.S.R., LXXII (72),3, pp. 555-8, 1950.

In 1947 V. N. Orekhovich and A. A. Tustanovsky (1) isolated a new crystalline albumin from the skins of animals, which they called procollagen, and studied its properties in detail (2-8). Aside from the purely biological problems which arose because of this discovery, the new albumin presented much interest from the standpoint of the physical-chemistry of macromolecules. By all external appearances it was similar to fibrous albumins, since, for instance, it gave a solution with high viscosity. On the other hand, a strong tendency to form crystals made it comparable to the globular albumins.

1. We investigated the procollagen from the skin of rats. The sedimentation constant S of the albumin was measured in the ultracentrifuge in concentrations of 0.045 to 0.45%. We also found the diffusion constant D at several concentrations, the specific volume and the viscosity. The experiments in the ultracentrifuge were carried on at 60,000 RPM, giving an acceleration of 250,000 g., and the sedimentation velocity was followed by the scale method of Lamma. The diffusion measurements were made on an apparatus provided with automatic recording devices over the period of two to three days.

For every experiment a fresh solution of procollagen was prepared in 0.325 citrate buffer at a pH of 3.3 (this pH is the most stable one for albumin). Afterwards the solution was filtered to remove a small amount of flocculent precipitate, using a No. 4 glass filter, upon which it appeared perfectly transparent. The percentage of procollagen was determined from the nitrogen content.



2. Who first thereogh inventigation by us of the properties of procollegen disclosed that of it did not become described, this protein was
really a concdisperse allowin, and therefore that it really belonged to the
group of globular proteins. Fig. 1 shows the sedimentation of procedinger
in the ultracentrifuge (albumin cons. of 0.21%). In Fig. 2 in presented
the variation of the sedimentation constant 3, normalized as usual to 20° 3,
and pure water, as a function of the concentration of the albumin. The increase of the sedimentation costant with a decrease in concentration is explained by the fact that large elengate molecules do not rove independently
of each other in concentrated adultions.

The measurement of the diffusion constant of procellagen presents contain difficulties, for them the albumin is maintained for some time at 25° C. It starts to denature and coegulate. Nevertheless, under high enough concentrations (above 0.3%) the error caused by this deformation is negligibly small and allows the calculation of the coefficient of diffusion with gratifying accuracy. In the concentration interval between 0.3 and 0.5% the diffusion constant has an average value of 2.2h x 10°7 cm²/tec. (adjusted also to pure water and 20° C.). One can accure that an average value of 1 occurs at the center of the interval, i.e. at 0.0%.

With S and D known it is possible to calculate the molecular usignt of the procollagen by the formula of Swedbergs

 (\mathbb{Z})

Here S is the sedimentation constant, i.e. the velocity of the macromolecules divided by the acceleration; A is the molecular weight; I is the specific velocity

C is the density of water at 20° C. (.3982); I is the diffusion constant;
T the absolute temperature (293° K.); and R the universal gas constant (8.313° 107 erg/degrees).

The specific volume was measured using a pycnometer and turned out (at 20° C) V = 0.720 ± 0.4005 .

By using the value at 0.1% conc. from Figure 2, S = (1.8 ± 0.05) x 10^{-13} one obtains for the molecular weight of procollagen M = $70,000 \pm 3500$. This allows one to compare procollagen to the large group of globular albumins near the weight of 70,000, which corresponds to four of the elementary units of Svedberg (17,500).

3. Now we consider the degree of asymmetry of the albumin. For this ourpose we use the value of S extrapolated back to zero concentration of albumin, i.e. $S = 3 \cdot 10^{-13}$ (Fig. 2). From the formula of sedimentations

$$S = \frac{M(1 - eV)}{f} \tag{2}$$

we find the frictional force f. This frictional force is related to the movement of the macromolecules in an infinitely dilute solution, i.e. when it does not depend on their interaction. If the macromolecules are spheres one can apply Stokes law $f_0 = 677$ ya, where a $\frac{1}{2}$ $\frac{1}{2}$ radius of the sphere (N is Avogadro's number).

For this case we make use of the empirically found relationships

(3)

If this equation (and it is exactly of the same magnitude as observed by Svedberg for the most asymmetric globular albumins) is completely explained by the asymmetry of the sphere it can be used to relate the formula for frictional forces to the dimensions of a prolate ellipsoid:

$$\begin{cases}
\frac{f}{f} = \frac{1}{A^{(1)}} \int_{\mathbb{R}^{2}} dx dx
\end{cases}$$
(4)

Using our data the ratio of two semi-axes becomes:

Consequently, the molecule of procollagen presents a cylinder with a length about 20 times its maximum diameter. Knowing the volume of the molecule $v = \frac{MV}{N}$, we find that the diameter $d = 16.7 \, \text{Å}$, and the length $L = 380 \, \text{Å}$.

Since the approximate amino acid composition of the macromolecule is known (8) we can calculate the average molecular weight of the residues, m=117, and therefore the degree of polymerization of the procollagen $\sqrt{s} \approx 600$. As the length of a single peptide bond equals about h Å, the length of the whole polymeptide chain in procollagen must be 2h00 Å, that is 6.25 times the length of the length L of the macromolecule. This undoubtedly means that the polypeptide chain is coiled in the macromolecule of the albumin.

4. In conclusion we will discuss the morphological reactions of a solution of procollagen under the action of salts. It was known that the presence of salt directly affects a solution of procollagen, changing, in particular, its viscosity. We examined in the ultracentrifuge a solution of procollagen in 3.32 M NaCl at a ph of 3.3.

The result was that next to the original peak of the procollagen appeared two new peaks, having a somewhat higher sedimentation rate (Fig. 3). This indicated the appearance of "dimers" because of the association of globules. Evidently this association can take place by different paths. The simplest variation of this association consists in the elongation of the globule by

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two (association in length). Using formulas (2) and (3) easily shows what will happen in this case.

Equation (3) may be written as:

For the dimer we have the corresponding

$$\left(\frac{f}{f_0}\right)_a = \frac{const (om)^{e/s}}{S_c}$$

Therefore

$$\frac{S_a}{S_c} = e^{2s_0} \left(\frac{S}{S_c}\right) / \left(\frac{S}{S_c}\right)_a$$

(5)

In our case of linear association $\binom{5}{5}$ = 2.78 (the length of the cylinder is doubled while the radius remains constant). This would give theoretically = 1.21. In actuality we obtained $S_2 = 3.083 \times 10^{-13}$, $S_1 = 2.558 \times 10^{-13}$, and therefore $S_2/S_1 = 1.20$, which constitutes an ideal correlation.

Definitely interesting is the second peak, a sedimentation with the values of $S_3 = 3.91 \times 10^{-13}$, $S_3/S_1 = 1.53$, which is almost equal to $2^{2/3}$ (1.59). This corresponds to an association of the two macromolecules along the length of the sides of a forming cylinder. In this case the semi-minor axis of the ellipsoid, as the length, stays constant, and the relation f_0/f_0 of the dimens is approximately equal to that of ordinary nolecules.

This experiment demonstrates that under the action of salts, as is known, the ionic attraction between primary groups of the albumin is decreased (at a pH of 3.0 the dissociation of the acid groups is mainly supressed). We also see that molecules of procellagen combine in pairs, forming two types of dimers - linear and lateral.

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Translated by Walter R. Stahl November 20, 1950

Chernikov, M. P. Amino acid content of ox procollagen. Doklady Akad. Nauk. U.S.S.R. 67, 345-7, 1949. (Presented by the Academician A. D. Speransky, May 5, 1949).

For the past three years U. N. Orekhovich and his collaborators have been conducting a thorough study of the procollagen (1-12) of proteins which were isolated in the laboratory in crystalline form from the skin and internal organs of humans and various animals.

Along with the study of the biological and physical chemical properties and the investigation of the prevalence of protein in the tissues of various animals (13,14) work was also conducted in the investigation of the qualitative (N. E. Plotnikov) and quantitative amino acid composition of the given protein. Some of the results of the chemical investigation of procollagen are set forth in this paper.

The procollagen was obtained from the hide of an ox immediately after skinning. The hide was washed, shaved, the fat and cellular tissues carefully removed, after which the hide was washed again in a current of running water and was ground in a meat chopper after having been cut up into small pieces. The resulting paste was used for the isolation of protein, globulin and procollagen.

After extracting the albumen and globulin with 1/15 M Na₂PO_{ψ} the paste was washed with a citrate buffer, pH = 3.5.

The procollagen was extracted with a triple quantity of citrate buffer in the course of 3-4 days at a temperature of 42°. Thymol and toluene were used in all cases for preservation.

The amorphous preparation of procollagen was obtained from the citrate extract by saturating it with dry NaCl up to 10%. The precipitate of albumen settled in the form of white flakes which were collected on the filter. To cleanse the procollagen from concomitant albumen, it was washed on the filter many times with a 5% solution of NaCl. To remove salts the procollagen was made turbid in a 15-20% solution of acetone with a subsequent centrifuging.

This operation was repeated to a negative reaction for chloride ions and a negative Millon's reaction, since the absence of tyrosine indicates the purity of procollagen compounds. Beyond that the protein was dried by the regular method with acetone and ether.

The resulting compound of the ox procollagen contained 0.47% ash, 17% nitrogen, 49% carbon and 7% hydrogen.

To determine tryptophane, tyrosine and phenylalanine, an alkaline hydrolysis was carried out with 5 N solution NaOH in the course of 6 hours by light boiling over a sand bath. The alkali was neutralized with a 14 N solution of H₂SO₄. The hydrolyzed substance was brought to a definite volume and then filtered.

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To determine the remainder of the amino acids an acid hydrolysis was carried out over a sand bath with 6 N solution of HCl in the course of 36 hours. The hydrolyzed substance was filtered after dilution. The hydrochloric acid was removed in a vacuum, after which the hydrolyzed substance was brought to a definite volume.

For the analysis of amino acids we used mainly specific methods which do not require any preliminary separation of amino acids. The methods were checked on hydrolyzed casein or gelatin. In all instances (except proline and oxyproline) we obtained fully satisfactory results,

Tryptophane (15, tyrosine (16), phenylalanine (17), methionine (18), cystine and cysteine (19), proline and oxyproline (20), were determined colorimetrically. Glycine and alanine by acidification with ninhydrin and the ensuing colorimetric determination of the resulting aldehydes (21,22). Arginine, histidine, lysine, amino succinic and glutamic amino acids were determined by the enzyme method (23,24) (in the laboratory of B. E. Zbarsky).

The determinations were made on a Schtufenphotometer or on a Specker absorptiometer.

The results of our investigations are tabulated in Table I. This table also shows the amino acid composition of collagen and gelatin (according to published data).

Table I

	In %	Dry Ash-free Pr	rotein
Amino Acids	Collagen (25)	Procollagen	Gelatin (26)
Tryptophane	0.4 4.28 5.8 5.8 0.3 11.3	0.0 0.0 2.3 9.2 4.6 2.9 5.2 11.0	0.0 0.44 2.2 8.0 4.1 0.79 6.7 11.5
Cystine) Cysteine) Methionine Glycine Alanine Proline Oxyproline	0.0 0.08 26.2 9.5 15.1 14.0	0,0 0,66 28.0 9.5 (20)	0.07 0.61* 25.5 8.7 19.7 14.4

^{*}In accordance with our data, in gelatin of the hide of a calf.

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By comparing procollagen and collagen in their amino acid compositions it can be seen that they contain equal quantities of lysine, glutamic acid, alanine, arginine, amino succinic acid, methionine and glycine and neither of the two contains tryptophane, cystine and cysteine.

Procollagen and collagen differ in their contents of phenylalnine, histidine, proline and oxypoline. Procollagen, unlike collagen, does not contain tyrosine.

Procollagen also differs in its amino acid composition from gelatin as is seen from the data presented in the above table. They contain different quantities of histidine, arginine, amino succinic acid, proline and oxygroline, and, furthermore, gelatin contains tyrosine while procollagen does not.

Thus, on the basis of our data, it can be said that procollagen in its amino acid composition is close to collagen and gelatin but is not identical to them and appears as an individual protein with connective tissues of the collagen type.

I am deeply grateful to Prof. V. N. Orekhovich for his valuable instructions and guidance in the performance of this work.

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Issued for publication on May 4, 1949

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Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5

THE AMINO ACID PROPERTIES OF BOVINE PROCOLLAGEN

M. P. Chernikov

(Reports of the Academy of Sciences of the USSR, LXVII, 2, p. 345-9)

V. N. Orekhovich and his associates have been conducting a multilateral investigation of the protein, procollagen, during the course of the last three years (1-12). They have obtained this substance from the skin and internal organs of both man and animals, using a product in crystalline form.

Together with the investigation of biological and physio-chemical properties, and also the distribution of the protein in question in different animals (13,14), there has been work done on the qualitative (N.F. Plotnikova) and quantitative amino acids properties of procollagen. Some of the results of the chemical studies of this protein follow.

Procollagen was obtained from cow hide freshly removed from the animal. The skin was washed, the covering of hair removed, the fat and subcutaneous cellular substance eliminated, and the hide then again washed in running tap water. The skin was ground up in a meat chopper after being cut into small sections. The mash so obtained was used to get albumins, globulins, and procollagen.

After the extraction of albumins and globulins with 1/15 M solution of Na₂HPO₁, the mash was washed with a citrate buffer of pH 3.5. The procollagen was removed with a three-fold soaking in citrate buffer over a period of 3-4 days at a temperature of 2°C. In all cases thymol and toluol were used as preservatives.

An amorphous preparation of procollagen is obtained by adding solid NaCl until a 10% solution was obtained. The protein precipitates as a white mass which is collected by filtration. In order to remove associated proteins, the filtrate is washed a number of times with 5% NaCl solution. In order to remove the salt, the procollagen was soaked in 15-20% acetone and subsequently centrifuged. This operation was repeated until one obtained a negative reaction for chloride ions and a negative Mellon's test, since the absence of tyrosine is a good criteria for purity of procollagen. Thereupon the protein was dried in the usual fashion with acetone and ether.

Procollagen prepared in this way has 0.47% ash, 17% nitrogen, 49% carbon, and 7% hydrogen.

For the determination of tryptophane, tyrosine and phenylanaline, the protein was hydrolyzed in 5N NaOH for six hours, with gentle boiling on a sandbath. The base was neutralized with a llN solution of sulfuric acid. The hydrolysate was then brought to a definite volume and filtered.

In order to determine the other amino acids, the procollagen was hydrolyzed in 6N HCl for 36 hours, with heating on a sand bath. After dilution the hydrolysate was filtered. The hydrochloric acid was removed in a vacuum and the volume adjusted to some definite amount.

In order to analyze the amino acids we used the accepted specific methods which do not require a preliminary separation of component amino acids. The process was checked with gelatin or casein. In all cases (with the exception of proline and oxyproline) we obtained fully satisfactory results.

Tryptophane (15), tyrosine (16), phenylalanine (17), methionine (18), cystine and cysteine (19), proline and oxyproline (20) were determined colormetrically. Glycine and alanine were found by colorometric measurement of aldehydes following oxidation with ninhydrin (21,22). Arginine, histidine, lysine, aspartic and glutamic acids were determined by enzymatic methods (23,24). (In the laboratory of B.E.Zbarckogo)

In Table I is presented the amino acid composition of procollagen with those of gelatin and collagen, obtained from literature. The figures are percentage of dry, salt-free protein weight.

	Collagen (25)	Procollagen	Gelatins(26)
Tryptophane	0.0%	0.0	0.0
Tyrosine	1.4	0.0	0.fr
Phenylalanine	4.2	2.3	2.2
Arginine	8.8	9.2	8.0
Lysina	4.5	4.6	4.7.
Histidine	0.8	2.9	0.79
Aspartic acid	6.3	5.2	6.7
Glutamic acid	11.3	11.0	11.5
Cystins) Cysteine)	0.0	0.0	0.07
Methionine	0.8	0.66	0.61 (out own data)
Glycine	26.2	28.0	25.5
Alanine	9.5	9.5	8.7
Proline	15.1	(20)	19.7
Oxyproline	14.0	(20)	Theh

Upon comparing collagen and procollagen according to amino acid composition we can see the following facts: both contain the same amount of lysine, glutaric acid, alanine, arginine, aspartic acid, methionine, and glycine and both do not contain any tryptophane, cystine or cysteine.

Procollagen and collagen differ in their contents of phenylalanine, histidine, proline and oxyproline. Procollagen, in distinction to collagen, does not contain any tyrosine.

A similar analysis can be done for gelatin. We thus see that though procollagen is similar to collagen and gelatin in amino acid composition, it is not identical. We may conclude it is a unique connective tissue protein.

At the Institute of Biological and Medical Chemistry Academy of Sciences of the U.S.S.R.

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Approved For Release 2003/01/29: CIA-RDP80-00926A005400050022-5

K. D. OREKHOVICH

(Akademiia Nüak USSR Doklady, 71:521, 1950)

THE PROCOLLAGEN CONTENT OF SKIN IN ANIMALS
OF DIFFERENT AGES

A new group of connective tissue proteins - procollagens, which were discovered by V. Orekhovich and A. Tustanovski, has been thoroughly investigated during the course of the last year (1-13). At the present time, we know the chemical composition, chemical and physio-chemical properties of procollagen. There has been much work done on the biological significance of this protein.

In this paper we present some observations based on the last series of experiments devoted to procollagen. The investigations deal with the content of procollagen in the skin of guinea pigs of various ages (from ten days to one year and more).

From the skin of normal, healthy animals we extracted procollagen by the method of A. Tustanovich (3) and determined the amount of this protein by comparing it with the weight of the dry, degreased skin; as well as by comparison to the mass of skin protein—collagen. The skin, removed immediately after killing the animal, was freed from hair, subcutaneous fat and cells, ground—up and treated by the methods given previously (204). The extraction of the protein was repeated five times. The procollagen thus obtained was dried to constant weight. The results are given in table 1.

As we see from table 1, guinea pigs at a young age (from 10 day to 5-6 months) have an amount of procollagen which varies between 7-10%. More mature animals (7-8 months) show a drop to 3-1%, while the old guinea pigs (8 months and older) have only 1-2% of procollagen.

Table 1

Age	Weight when killed (g)	% Pro- collagen	Age	Weight	% Pro- collagen
10 days 20 # 2-6 months "" " " " " " " " " " " " " " " " " "	80-117 120-170 262 287 3347 323 347 323 367 377 384 400 402 448 454 535	67.303554804540352 1064540352	7-8 months " " " " " " " " " " " " " " " " " "	690 670 7150 760 760 840 840 862 975 984 1050	4 2 2 3 4 1 1 3 1 3 6 2 1 3

In order to answer the objection that our data depends on the different extraction conditions optimum for older and younger animals, we conducted the following experiment.

We took the skin from five animals weighing 300-400 grams and from two guinea pigs of 725 and 827 g. The mash obtained from these two groups of animals was divided into eight portions. Each aliquot was covered with buffer solution of the following oH: 1.5, 2.0, 3.6, 3.95, 4.12, 4.47, 5.02. Procollagen was obtained from the extracts. We found (see Table 2) that the results were the same regardless of the extraction used, and that they also corresponded to earlier conclusions, i.e., that the skin of older enimals has several times less procollagen than that of younger ones, although the optimum extraction conditions vary for the two groups.

On the basis of the facts presented above, we can be confident of the fact that as an animal gets older the percentage of procollagen in its skin decreases. It is notable, that in old animals the process of formation of new collagen fibers is also slower. There is reason to suspect that there is some relationship between the formation of collagen and the concentration of procolagen within the skin.

Table 2

THE AMOUNT OF PROCOLLAGEN OBTAINED FROM THE SKINS GUINEA PIGS UNDER DIFFERENT EXTRACTION CONDITIONS

Weight (g)				pH sol	Lution). TO	11 117	E 02
Weight (g) 300-400 725, 827	1.5 11.6 2.1	2.0 14.3 3.6	3,0 17.1 3.3	12.3 .3.6	4.2 8.8 9.8	7.2 4.4	12.0	8.8

Institute of Biological and Medical Chemistry of the Academy of Medical Science of the USSR

Presented by: 7/1/1949

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Same as previous article

Translated by: Walter Stahl 4/25/1951 Orekhovich, V.N. et al. The procollagen of hide. Blokhimiya 13(1), 55-60 (1948). (Translated from Russian by Mrs. M. Baker)

We have isolated and obtained out of the hides of animals, albumin in crystalline form which evidently belongs to a special group of connective tissue albumin. There are some reasons to assume that it is a biochemical predecessor of collagen, and we therefore named it procollagen (1-4). In this article are given the data of the prevalence of procollagen in the animal world and some facts about the mature of this albumin.

Material and Methods

For the purpose of experimentation animals of various classes of vertebrates and invertebrates were taken. Of the mammals rabbits, rate, calves, dogs and cats were taken; of birds - hens; of reptiles - turtles and grass snakes; of amphibians - frogs; of fish - pike-perch. Besides that, tests were made (for content of procollagen) with human skin and tissues of various species of invertebrate animals.

The hides of animals were taken immediately following their slaughter and the removal of their blood. The outer coverings (wool, scales, feather and horny membranes), hypodermic tissue, fat, etc. were carefully removed and the remaining material was ground up. The resulting paste was used for the purpose of isolation of albumins, globulins and procollagen (crystalline and amorphous).

The albumins and globulins of the animal hides were isolated from the paste by means of five times the volume 0.3 M Na₂HPO_{μ} within 24-36 hrs. at *2°. The globulins were isolated by adding an equal volume of a saturated solution of (NH $_{\mu}$)₂SO $_{\mu}$ ° By adding (NH $_{\mu}$)₂SO $_{\mu}$ to saturation, the albumins were isolated from the filtrate.

The crystalline procollager was isolated by the following method. After extraction of the albumins and globulins by means of the phosphate, the remaining paste was washed once or twice with small batches of citrate buffer pH = 4.0. To the washed out paste was added citrate buffer (pH = 4.0) five times the amount by volume (the weight of the paste). This mixture was left at a temperature of (+1°, +2° C) for 24-36 hours. After filtration a transparent viscous solution containing procollagen was obtained. This filtrate was put in collodion or cellophane bags for dialysis under running water or 0.01 M Ma2HPO4. In 24 hours crystals of procollagen in the shape of long needles were precipitated. The crystalline procollagen was collected on a filter and dried with filter paper. The moist crystals were kept cool. To obtain dry albumin the crystalline procollagen was desiccated with alcohol and ether and then dried to a constant weight at 104-105° or the

2.

dry crystals were obtained by the method previously published by us (1).

Amorphous preparation of procollagen we obtained by the following method. To the citrate extract of the hide paste was added an equal volume of a 10% solution of sodium chloride. The procollagen was isolated from the solution. The precipitate of albumin was collected on the filter and washed several times with a 4-5% solution of sodium chloride. The moist procollagen was kept in the cold.

The ultraviolet spectra of the absorption of the albumin solutions were studied with the aid of a Smith and Hensh quartz spectrograph (medium dispersion) by the method of the threshold of blackening. The source of light was a hydrogen lamp of low voltage of the GOI system.

The procollagen was boiled with crystalline pensin (2), trypsin and chymotrypsin (prepared by method of Kunitz and Northrop) and preparations of papain and eatersin (an acid glycerine extract from the liver of a rabbit)

Results of Investigations

The Occurrence of Procollagen

We encounter procedlagen in the skins of all classes of vertebrate animals.

a. Mammals. We found procollagen in the skin of all the animals we examined (rats, rabbits, calves, dogs and cats). The amount as well as the conditions of its extraction and crystallization vary with different animals. The age of the animal is evidently important. The younger the animal the easier it is to extract the albumin and the easier it crystallizes.

The skin of rabbits contains about 4% of procollagem (to the dry weight of the skin). The albumin crystallizes out of the extracts more completely and much faster with a citrate buffer solution of an initial pH = 3.8. The crystals are in the shape of needles with a maximum length of 260 µ (see diag. 1). Out of the skin of a dog the procollagen is isolated in a much lesser amount, namely 0.6%. The crystals form during dialysis from the extracts with citrate and buffer solutions with an initial pH of from 3.0 to 4.0. The maximum length of the crystals is 130 µ. Crystal procollagen was also isolated from the skin of a cat. The maximum length of the crystals is 130 µ. The albumin crystallizes much faster in the process of dialysis under 0.01 M Na₂PO₄, the crystallization proceeds much slower in the dialysis under running water. Procollagen becomes isolated very easily

out of the hide of a newly born calf. The albumin is crystallized from the extracts with citrate buffer solutions with an initial pH of from 2.0 to 5.4. The yield of albumin is 0.86%. The maximum length of crystal is 195 μ (see diag. 2)

- b. Birds. The albumin was isolated from the skin of chicks about 1.5 months old. The crystallized albumin is precipitated from the extracts with buffer solutions of an initial pH 2.7-5.5
- c. Reotiles. The albumin was isolated from the skin of a Middle Asiatic desert turtle. The procollagen of this reotile discloses the peculiarity that the albumin crystals can be obtained only if the pH of the citrate buffer solution (used for extraction) does not exceed 3.0 and the dialysis is carried out under 0.01 M Na₂HPO₄ but not under running water. The maximum length of crystals is 52 μ . So far we have not succeeded in isolating the procollagen from the skin of a grass snake.
- d. Amohibians. Crystalline procollagen in only very small quantities can be obtained from the skin of frogs (Rana temporaria). The dialysis of the extracts is better under a solution of Na2PO $_{\mu}$. Maximum length of crystals is 65 μ .
- e. Figh. The skin of fish, seemingly, contains the largest amount of procollagen. 2.5% of procollagen (dry albumin with dry weight of skin) was obtained from the skin of a pike-perch. The albumin passes into the solution very easily and crystallizes in the dislysis under running water as easily as under 0.01 M Na₂HPO₄. The albumin crystals precipitate from all the extracts with citrate buffer solutions with a pH of from 3.5 to 5.5. The maximum length of crystals is 90 µ. So far we failed to isolate the procollagen from the tissues of the invertebrates. It is possible that the modes of extraction of this albumin have to be different from the ones developed by us in the case of vertebrates.

The Nature of Procollagen

Procollagen is a globular albumin soluble in acidified water and not soluble in neutral and weak alkaline medium. In concentrated solutions it has a very high viscosity. It precipitates from solution even in the presence of small concentration of sodium chloride (5%) and other neutral salts. In amino acidic composition it approaches collagen and gelatin, but in solubility and salting out and other properties it differs from them. During boiling of the solutions or during suspension, procollagen turns to gelatin.

Below we will describe certain facts which characterize the nature of this albumin.

3.

4.

- a. <u>Ultraviolet Abscrotion Spectra</u>. We have made a study of the ultraviolet absorption spectra of solutions of procollagen, albumin and globuline of skin and gelatin.
 - Diag. 3. Ultraviolet absorption spectra of solution of albumin and globulin of skin. The unbroken line is that of albumin and the dotted line is that of globulin.

For analysis we precised 1, 3, 4, 6 and 10% acidified water solutions of procollagen (pH of solutions from 2 to 5), 1, 3, 5, 6 and 10% water solutions of gelatin from the hide of a calf (pH of solutions from 2 to 7). Since the majority of the above indicated solutions at room temperature form a jelly, it was necessary to heat the solutions to 30-35° in order to fill the cuvette of the spectrograph. Albumins and globulins of the skin were used in the form of 0.75% salt solutions of albumins (pH of solution - 6.5-7.0).

Solutions of albumins and globulins of skin give a typical albumin spectrum with a maximum absorption of about 2,800 A(see diag. 3) which is defined in these albumins by the presence of considerable quantities of tyrosine and tryptochane. In the spectra of procollagen solutions, the characteristic maximal absorption is absent and only in the spectra of concentrated (over 3%) solutions of albumin there are 5 absorption lines, three of which have a width of 30 A each and 2-20 A each. These lines lie between 2570 and 2,600 A; 2,640-2,660 A; 2,690-2,710 A; 2,760-2,790 A; 2,840-2,870 A (see diag. 4).

In the spectra of 6% and 10% solutions of gelatin obtained from the hide of a calf, we succeeded in detecting only 3 absorption lines 2,570-2,600 Å, 2,640-2,660 Å and 2,650-2,700 Å (diag 5). Comparing the absorption spectra of these two albumins it is seen that the absorption lines for gelatin noted by us are identical with the corresponding lines in the absorption spectra of procollagen. The next 2 lines of absorption which lie in the longer wave section of the ultraviolet we could not detect in the spectra of the gelatin solutions. As is known (5), the characteristics of absorption spectra of gelatin are dependent upon the absence from the albumin of tyrosine and tryptophane and on the presence of phenylalanine. On the basis of the above-mentioned facts we can assert that phenylalanine is also present in procollagen. For the time being we cannot explain on what the characteristic of the absorption spectrum of procollagen depends.

b. The Relationship of Procollagen to Proteinases. We made a study of the digestibility of crystalline procollagen with various proteolytic enzymes. It became clear that this albumin is well digested by tissue proteinases (cateosin and papain) and comparatively poorly by the proteinases of the digestive system

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(pensin, trypsin and chymotrypsin) (see table below)

TABLE I

The Intensity of Fermentative Hydrolysis of Procollagen at Various pH Media

g	Increa	se of	Amino	Nitros	en in	24 hrs	, in	specime	n (in	mg.)
Enzymes					ga	[
	2.3	3.2	3.67	4.0	4,4	4.5	4.9	5,60	6.0	7.8
papain	1.30	1.40	2,10	0.90	1.50	ब्यून स्टब्स	1.50	का संद	1,60	477 ago
cateosin	0,0	0.70	1,90	2.30	cm +40	2°50	an ⇔	1.60	1.10	Carriages
persin	0.0	0 , 50	0,90	0.0		0,60		0.30	sas at	angli distr
tryosin		~ · · ·	on sen	40 co			. ACD - ACD	****	a» =3	1.00
chymotryo sin		- FT	or se	635 Ho	جه نبت <u>.</u>	est) (800)	es2 #11		gis wo	0.60

In the process of hydrolysis of albumin in 24 hours here is liberated in the form of free amino nitrogen, the following percentages of the entire albuminous nitrogen: with catepain - 40%; under action of papain about 30%, by hydrolysis with trypsin only 15%; with persin 14%; and with chymotrypsin 9%. Denaturing of albumin by boiling has no effect on the extent of its hydrolysis with trypsin and chymotrypsin.

Chemical Composition of Albumin

We have already published certain data about the elementary composition of procollagen from the skins of rats and rabbits. In highly purified and multiprecipitated procollagen there is: carbon 49%, nitrogen 16%, hydrogen 7.5% (averages).

A small amount of phosphorus (0.15%) which is found in procellagen we are inclined to attribute to impurities which are difficult to isolate by methods available to us. On the basis of results from the ultraviolet spectrography we can already conclude that procellagen definitely or almost definitely does not contain tyrosine or tryptophane. On the basis of chemical reactions we have established that there is no tyrosine (Millon's reaction negative) and only traces of tryptophane (xanthoproteic reaction and voisene reaction). Reactions for sulphur are negative. We isolated from albumin histidine (in the form of nitroanilide), arginine (in the form of flaviamide) and lysine (in the form of nitroanilide). The general content of diamino acids about 5%.

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Procollagen contains over 10% of dicarboxylic acids (isolated by us in the form of barium salts). Albumin contains a considerable amount of proline and hydroxyproline. The reaction for carbohydrate is positive.

Conclusions

The albumin of skin which according to our tests possesses a high biochemical instability when isolated in crystalline form, differs in a series of characteristics from heretofore known albumins. It may be assumed that it is a forerunner of collagen and is, therefore, named procollagen by us.

It seems to us that there is basis to classify this albumin in a special group of connective tissue albumin and that procollagen will not be the only representative of this group. Procollagen is widely spread in the animal world and it is safe to say that there are not any species of vertebrates which do not contain this albumin. According to our tests, procollagen is present not only in the skin but also in a series of other tissues and organs of animals. In particular, it was possible to isolate from the tendons of a bull a crystalline albumin which resembled the procollagen of skin. It will be of interest to compare the identity of the procollagen isolated by us with the crystals of albumin in compounds of the sinews of a rat's tail as observed under a microscope by Nageotte and with the amorphous compound of collagen like albumin isolated from the skin.

We consider it our duty to express our thanks to our collaborator K, D. Leontgeva of the physiochemical laboratory for her great help in spectrography

Titles of Diagrams

- Diag. 1. Crystalline procollagen from the skin of a rabbit.
- Diag. 2. Crystalline procollagen from the hide of a calf.
- Diag. 3. (explained in text)
- Diag. 4. Ultraviolet absorption spectrum, 4% solution of pro-
- Diag. 5. Ultraviolet absorption spectrum, 6% solution of gelatin

7.

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Approved For Release 2003/01/29: CIA-RDP80-00926A005400050022-5

V. N. Orekhovich, A. S. Konikova, K. D. Crekhovich
and N. N. Dobbert

CONCERNING THE METABOLIC TURNOVER RATES OF VARIOUS ORGAN AND TISSUE PROTEINS

Akademiia Nauk SSR Doklady, 71:105, 1950

It is well known that there is a continuous renewal of the parts of an organism; that its proteins, fats and carbohydrates do not remain unchanged after their synthesis and inclusion in organs or tissues, but are constantly being turned-over. Naturally, the study of this process is a very essential and important part of the deciphering of the functions of the organism. This is especially interesting from the standpoint of proteins.

For our studies we have made use of tagged atoms to determine the turnover rates of different organ and tissue proteins and their separate fractions. This problem has been studied by Schoenheimer and his associates (1), who made use of tagged amino acids. They only investigated the time for inclusion of these substances into different proteins, but not the overall turnover rate of the protein. Ussing and Krough? investigated the overall renewal of proteins with the aid of a heavy hydrogen isotope (deuterium), carried in the form of heavy water. They were only interested in the turnover rates of proteins of the skin, muscles and some internal organs.

We also made use of heavy water, studying the turnover rate of proteins from the organs and tissues of white rats, as well as some of their constituent fractions.

Heavy water was given to the animals at such a rate that after a few days its concentration in the body reached 1%. Then the animals were killed and proteins extracted from all organs and tissues. The proteins were carefully treated to remove physically bound water, dried to constant weight and incinerated. After being suitably cleaned, the water formed was analyzed for atom per cent deuterium by the flotation density method.

The data we obtained are presented in Table 1. It is seen that the fastest turnover rate occurs in the proteins of the liver, while that of the skin and muscles is the slowest. The rest of the investigated proteins lie between these limits.

We tried to characterize the turnover rate of proteins not only with respect to how fast they took up deuterium, but also as to how fast they lost it again. With this goal we worked with a group of rats which were given heavy water for twelve days and then not killed immediately, but after twelve more days had elapsed.

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Table 1
TURNOVER RATES OF PROTEINS OBTAINED FROM VARIOUS ORGANS

Organ	Excess of deuterium (at %) after intro- duction of D O	Turnover rate (%) of pro-teins	Excess of deuter- ium (at %) after stopping introduc- tion of D ₂ O	Differ- ence (%)
Liver	0°535	23.3	0.109	0.123
Intestine Spleen Kidneys Stomach Heart Lungs Brain	0.170 0.167 0.162 0.137 0.136 0.115 0.112	17.0 16.7 16.2 13.7 13.6 11.5	0.094 0.107 0.079 0.084 0.106 0.078 0.092	0.076 0.060 0.083 0.053 0.030 0.037 0.020

As we see from the data presented in Table 2, the rate of loss of deuterium from proteins of organs and tissues is not the same as their rate of uptake. For example, the uptake of deuterium by the kidneys ranks fifth, while it is second as far as rate of loss is concerned.

Table 2

LOWERING OF THE AMOUNT OF DEUTERIUM IN PROTEINS OF VARIOUS ORGANS 12 DAYS AFTER IT CEASED BEING PRESENTED (%)

ORGAN	% DECREASE
Liver Kidneys Intestine Stomach Lungs Soleen	53 54 40 38 31 30
Heart	20
Brain	1.7

There is the same lack of correspondence in the uptake and loss rates of the proteins from lungs and stomach. This is explained apparently, by a difference in intensity of exchange rate of structural moleties between various organs. The disproportionally low figure for kidneys can be explained not only by the intensity of turnover of amino acids containing deuterium, but also, apparently, because of the dilution of their own structural proteins with those from organs which contain a low concentration of the isotope.

Together with the study of the intensity of the turnover rate of the proteins from different organs and tissues, we also study the turnover of various protein fractions derived from the same organ. We anticipated that by this method we would be able to answer the question, to what extApproved For Release 2003/67/29? ETA-RDP80-00926/7003460050022-5 f a given

organ is specific for individual fractions which go into the makeup of the organ. The figures we obtained are presented in Table 3.

Table 3

INTENSITY OF TURNOVER OF VARIOUS PROTEIN FRACTIONS
FROM DIFFERENT ORGANS

Protein	Excess of deuterium (at %)	Turnover rate (%)
Blood proteins Blood globulins Liver glovulins Skin globulins Albumins and globulins of the skin Collagen of the skin	0.181 0.148 0.137 0.138 0.177	18.1 14.8 13.7 13.8 17.7
Procollagen of the skin Ossein (collagen ?) Muscle proteins Myogen	0.123 0.115 0.101 0.089	15.5 15.5 10.1 *,0

As can be seen from this data, the turnover rate of globulins of the blood, liver and skin is practically the same, while the intensity of turnover of the collagen of the skin is much lower than the skin globulins. In this way we can see that the turnover rate of the fractions varies in each given organ.

Besides examining the turnover rates in normal adult rate, we also studied them in newborn animals and their mothers. For these experiments we fed a pregnant rat heavy water for twelve days before the birth of her litter. The same day that the rats were born they were killed, together with their mothers, and proteins from different organs analyzed for their content of deuterium. The results are given in Table 4.

Table 4

INTENSITY OF TURNOVER IN ATOM PERCENT OF DEUTERIUM

Tissue	Excess Mother	atom-% Newborn	Protein to Mother	rnover 4 <u>Newborn</u>
Skin Internal	0.070 0.187	0.236 0.254	7.0 18.7	52°6
organs Carcass Muscles Head	0.070	0.264 0.252	7.0	26.4 25.2

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As we can see from Table 4, the turnover rate of the oroteins of the skin and muscles of the mother is markedly less than that of normal adult rats (Table 1). This indicates that there is a sharp decrease in turnover rate of muscle and skin proteins during organancy. The turnover rate for the internal organs of the mother is normal. The proteins of all the tissues of the new born animals show almost the same excess of deuterium, which is much higher than any of the rates of the tissues of the mother. This may be explained, perhaps by the fact that all the organs and tissues of the embryo are being built up of free amino acids, without its making use of the formed proteins of its mother's organism.

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Presented: 6/31/40

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A crystalline protein resembling collagen is obtained from rabbit or rat skin by the following procedure: Shredded skin, freed from muscle, porous epidermis and superficial fat, is extracted with 0.1 M buffer composed of completely or partially neutralized organic acid (citric, succinic, oxalic, lactic, tartaric, adipic or glutamic). The optimum pH for extraction varies with the type of skin and the organic acid employed. In the extraction of rat skin the optimum pH is 3.5 for oxalic acid buffer and 4.1 for citric acid buffer, while in the extraction of rabbit skin with citric acid buffer the optimum pH is 5.7. Extraction with 5-6 ml. of buffer/g. of skin requires 12 hours under mechanical agitation. Extraction and subsequent operations are carried out at 7-5 C. The buffer extract is clarified by centrifugation and filtration. The filtrate is dialyzed against water, pH 7-8. When the buffer concentration falls to below 0.01 M, crystal separation occurs spontaneously. Crystals are predominantly needleartion of the protein of the protein and filtration of the protein of the protein are skin yields 2.7 g. of dry crystallina protein. M. C. Brockmann

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A. L. Zaldes and S. L. Puoko (Akademiia Nauk USSR Doklady, Vol. 73: 991, 1950)

ELECTRON-MICROSCOPIC INVESTIGATION OF THE EFFECTS OF ALKALIS AND PANCREATIN ON COLLAGEN

The electron microscopic examination of collagen has disclosed the presence a regularly repeating structural pattern in the fiber (1).

There is a strong interest in the effects of various treatments on this structure. Since the production of collagen by the skin involves the presence of specific fluids and enzymes of the pancreas—pancreatin, we decided to study the influence of these reagents first, especially since the illucidation of the action of lime on collagen is inadequate (2), and the literature on the enzymatic treatment is totally lacking.

Pieces of collagen, obtained from the frontal part of a steer skin, cut in a size 2 x 10 cm. were processed with a solution of lime-water containing calcium hydroxide in a concentration of 10g/l. They were kept in this suspension over a varying period of time, ranging from four days to two years. Before examination, the swollen fibers were neutralized in 5% bisulfite and carefully washed.

The enzymatic treatment of fibers kept in alkali for four days and then neutralized was carried out with a one per cent solution of pancreatin at a temperature of 37°C, for periods of 3 hours, 8 hours, 12 hours, and four days, with constant asitation all the while. With the protracted pancreatin treatment, we used a daily change of pancreatin solution.

Then the collagen was washed with water and dried in etherelcohol.

From the prepared tissues we cut sections in a freezing microtome; they were then dispersed for five minutes in a magneto-strictive apparatus with a frequency of about & kilohertz.

The dispersed collagen was placed on a collodion film and shadowed with chromium for increased contrast, using an angle of $15^{\rm Q}$.

The results show that treatment with limewater over a period of four days causes absolutely no change in the structure of the collagen (Fig. la). With collagen soaked for a month we begin to see disruption of the structure, which becomes more marked upon two months' processing. Together with regions without local changes, we can see strongly deformed fibrils, mixed up among each other and with a loss of periodic structure (Fig. l,b).

A complete lack of regular structure is seen in collagen after two years of treatment with calcium hydroxide solution (Fig. 1,c).

We thus conclude that the limewater has to act a long time to cause complete breakdown of structure; its action is slow and irregular.

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A similarly unpredictable change in form is caused by the action of pancreatin. But the character of the alteration is quite different.

Enzymatic treatment carried out over a period of three hours does not change the typical electron-microscopic picture of collagen.

It is interesting to note that the indicated duration of treatment corresponds to collagen produced in a certain way, and that our experiments under these conditions showed absolutely no alteration in the finest details we could resolve.

In proportion to extended treatment in enzymatic solution, a destruction of the collagen starrs to appear gradually. In the e.m. picture of material processed for 8 hours, we can see badly damaged regions together with ones that show clearly defined striations (Fig. 2,a). Treatment over the course of 12 hours causes partial unraveling and breakdown of the fibers (Fig. 2,b), while four-day digestion yields material with no regularity at all, giving just "debris" under the electron-microscope (Fig. 2,c).

In one of our papers (3), we noted that electron-micrographs of replicas of all collagen fibrils gave a definite orientation of perpendicular bands. Similarly with dispersed fibers, we noted this regularity of structure in all fibrils which could be observed close enough to each other. We can see in Fig. 3 that the dark and light regions correspond in adjacent fibers. This is an adequate proof of the organized spatial arrangements of the molecular micelles found in collagen.

We are indebted to A. N. Mikhailov and A. I. Frimer for their interest and work.

Central Scientific-examining Institute of the Leather-wear Industries of the USSR

Presented: 6/10/1950

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Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5

DIAGRAM CAPTIONS

- Fig. 1 Collagen treated with milk of lime, dispersed with sound and shadowed with chromium. Length of treatment: a-1 days, b-2 months, c-(beta)-2 years
- Fig. 2 Collagen after enzymatic treatment, dispersed with sound and shadowed with chromium. Length of enzymatic treatment: a-8 hours, b-12 hours, c-4 days
- Fig. 3 Ultrastructure of fibrils, dispersed with sound and shadowed with chromium

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Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5

Zaqdes, A. L. and S. L. Poxpko. The electron microscopic examination of collagen. Reports of the Acad. of Sci. of USSR LXV, 2, pp. 227-8, 1949. From the Central Research Inst. of the Leather-footwear Industries.

The electron microscope examination of the primary fibrous protein of the skin, collagen, is of interest from two viewpoints. It is, of course, allowing the growth of ideas about collagen itself and its production by the skin. But it is also shedding light on the nature of fibrous proteins, in general, which are more widely distributed.

To the present time x-ray analysis has been used to examine the fine structure of collagen. This did not allow, however, the examination of the form and inter-relation of the elements of the subfibrils whose thickness ranges from 100-50 A.

For the electron microscopic examination of collagen we used both direct observation and replica technique. Our exmination by the latter method confirmed the earlier observations of the structure of this material (1).

A more thorough method of direct observation was the result of sectioning in a microtome to a thickness of a few microns and then dispersing with ultrasonic vibrations of frequency of about 8,000 c.p.s. for 3-5 minutes. The preparations so obtained were deposited on films of collodion or aluminum oxide and shadowed with chromium to give more contrast (2).

Untreated fibers were examined, as well as those stained with various salts of heavy metals and extracts from wood of the oak.

All examinations of the fibers have revealed periodic bands which are arranged perpendicular to the long axis or as a spiral. The spiral form appeared very sharply in some of the photomicrographs,

When different forms of treatment are used, the width of the striations changes, revealing still finer bands within. Thus, material treated with tan-bark extracts shows an over-all spacing of about 700 A and a band width of about 170 A.

The polysterol-quartz replica of untreated collagen fully confirms the above findings (2).

Similar results are also obtained through the use of methyl-methacrylate-quartz replicas which avoid the necessity of heating the object during preparation.

Electron microscopic examinations of collagen fibers by American authors (3) gives the following picture of its structure. The wide bands of the collagen subfibrils consist of several smaller bands which show a characteristic, periodic density. The bands extend the width of the whole fiber. This conclusion is

not in agreement with our own experimental results.

If the subfibril really has the chain form given above, replicas of individual fibers should show the form of the fiber without its internal structure. Electron photomicrographs prepared by us, however, show a periodic structure consisting of two mutually perpendicular spirals of subfibrils.

These observations bring us to the conclusion that the molecular components of collagen show a contour structure.

The accompanying electron microphotographs (none) of untreated, shadowed collagen fibers (chromium) show the outer structure of the material. The shadows show the same periodic structure, proving the presence of the relief.

The observation of fibers at the moment of bursting due to the action of the electron beam on the fiber or supporting film has confirmed the above conclusion. The break always occurs at the lightest, and then the most thin, bands. The more dark, and therefore dense, bands then take on the form of an extended droplet. In the broken fibers the spacing between the light and dark bands is still more apparent, supporting the above hypothesis.

It is interesting to note that the broken fiber takes on a more uniform appearance when stained with phosphotungstic acid, apparently due to a more uniform absorption of the metal.

Received: Dec. 30, 1948

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Diagrams

Unraveled collagen Fig. 1.

Native collagen; 6-stained with phosphotungstic acid; Fig. 2a. Y-oak extract treated, P-uranyl acetate stained;

Chromium shadowed collagen fiber; 5-polysterol quartz Fig. 3A.

replica of collagen

PTA treated fiber distorted by bursting of underlying film. Fig. 4.

Translated by Walter Stahl

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Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5

(Akademila Nauk USSR Doklady, Vol. 73:379, 1950)

THE TLEGTRON-MIGROSCOPIC EXAMINATION OF TOLLAGEN USING THE REPLICA TECHNIQUE

The orimary advantage of the replica technique for examining collagen (1) lies in the fact that it does not require that the fibers be treated in any structurally-disturbing way (for instance, as in dispersion by sonic vibrations) and that one can obtain the impression of a block of material, according to which it is possible to judge the architecture of the tissues. The investigations of collagen by the collodion replica technique available in the literature (2) deal with samples of the fibers first dispersed by various methods (sonic vibration, colloid mill, mechanical unraveling with needles). By this method one obtains imprints of single fibers only, since by such treatment the histological structure of the collagen is destroyed.

Together with the study of dry objects by the use of polysterol-quartz replicas, we have worked out a method of getting replicas of wet materials. To achieve this goal we heat an aliquot of methyl methacrylate monomer in a reflex condenser, adding the polymerization catalyst. This is carried on until the liquid becomes slightly viscous. Then the top layer is cut off a piece of collagen in a freezing microtome and the remaining material is submerged in water.

After recovering the block from the vater, the surface chosen for examination is freed from excess water by using a piece of filter paper for drying (this process must be carried out rapidly). On a surface prepared in this way is poured the freshly prepared and slightly polymerized methyl methacrylate. In order to prevent drying, the specimen is covered with the liquid on all sides. The consequent polymerization process takes place on the object itself. The first layer of thin film is reinforced by a secondary coating of polymer until a thickness is built up which makes it convenient to remove the film (about 0.1 mm). Then the film is taken off and it is dusted with quertz in a vacuum to give a thin film (order of 100-300 A), apolied on the side facing the collagen. The quertz film is freed from the methacrylate by dissolving the latter away with dichlorenthane.

Using the replica method we examined soecimens of collagen which were untreated and some which we prepared in various ways: in calcium hydroxide, enzymes (pancreatin), salts of heavy metals (PTA, PMA, and uranyl acetate), fixed in formaldehyde, primary sulfate of chromium and woody oak extract. In all cases we observed the presence of regular perpendicular bands of different densities (see figs. 1,a,b; 2,a,b; 3,a,b).

We were able to observe a greater differentiation on the replicas from wet fibers than on corresponding dried ones. Thus, in the case of the chrome-tanned material, besides the 63½ A period there was one of 297 A. By ordinary observation in the electron microscope we observed roughly the same period (282 A) in collagen fibrils. In some of the replicas, obtained from the specimens treated with extract of the woody oak, we were able to see spherical particles lying on the fibrils (see fig. 4). In all probability, these particles are composed of tanning, not part of the collagen but only adsorbed on its surface.

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Vith an accuracy of about \$15 A. The periodicity is observed to the limits of each method of preparation (see table 1). We can divide the observations on the periods into three groups: 1-with a repeat period of 640 A in the wet state; 2-the period of dry, untanned objects (here too are specimens treated with PTA and formaldehyde) which ranges from 540 - 560 A; 3-a period for fibers treated with various tanning agents and dried, which is about 430 A.

<u> Table 1</u>

REPEAT PURIOD OF THE FIBER

Form of treatment	Repeat Period in A	Fiber Width in A
USING POLYSTEROL-QUARTZ REP	PLICAS OF DRY FIBE	CRS
Untreated Calcium hydroxide (3 days) " " (1 month) Pancreatin (3 hours) " (12 hours) P.T.A. Formaldehyde P.M.A. Uranyl acetate Primary sulfate of chrome	575 5746 575 546 558 409 409	650 980 820 1140 910 990 800 705
USING METHYL METACRYLATE-QUA OBJECT	RTZ REPLICAS ON W	ET
Calcium hydroxide (3 days) Pancreatin (8 hours) Pancreatin (12 hours) Primary sulfate of chrome Extract of woody oak Formaldehyde	640 640 625 634 651 609	1200 1100 1100

Replicas have not only allowed us to illucidate the repeat periods of fibrils, but also the relative constancy of their width. For the various forms of collagen treatment the average width is about 0.1 microns. In the case of the wet fibers, tanned with the primary sulfate of chromium and formaldehyde, calculations show an increase in width of about 1020%.

Replicas of all the examined fibers show a definite orientation of the bands in different fibrils; the dark bands as well as the light coincide. This fact, and also the constant period of repetition, is characteristic of the collagen structure, indicating a definite soatial arrangement of molecular groups.

In conclusion, we are obligated to express our thanks to A. N. Mikhailov and A. I. Firmer for their interest and work.

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Presented: 6/14/1950

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	7 :	53	36	426.	:	55	37	হ'ত	:	54	36	es.	:	74	57	-	:	86	61	-	:	93	5 6	-	:
	8 ;	54	l _t l _t	-	:	54	37 36 36	=-	2	58	30		:	71	45	~	:	89	55	-	:	95	59 61	-	:
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	4)	50	29	0.10	:	25	<i>3</i> 3 35 36 36	•••		62 60	36	***	:	80 70	57	-	:	82	54	-	:	37	61 61	-	•
	34 1	50 53	39 28	2 50	:	55 60	25	440	:	56	35 38	0 15	:	69	59	-	:	74	49 46	-	:	90	64	-	:
	16 :	50	36	0,50	•		၁၀	C P	2	50 50	3 6	0 10	:	71	45 60	~	:	73	•	-	:	93	61	_	•
	47	53	ار 54	U. 15	:	57 57	200		3	55 55	36		:	66	56	***	:	74 73	50 49	_	:	9 7 90	56		:
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	20 :	50	39 39	U - Z :)	;	61	37	_		61	29	12, 00	ì	87	50	_		36	50	_	:	94	57	_	:
	21 :	50	34	-	i	59	36	_	e e	65	35	**	:	91	55		:	95	55	_	:	97	58	_	:
	50 1	50	39		•	61 <u>.</u>	Į, j	0.10		63	\tilde{u}	W		91	55		:	94	52		:	9 5	61	_	· ka
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	24 :	50	39	0, 25		54	33	199		70	100	He		81	52	~	:	99	61	-	,	101	63	_	. 🖺
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	27 :	48	37	95	•	43	2/4	14	1	71	1,2	4-	4	94	62	176	;	51	56	œ.	:	106	63	-	:
	28 -	54	52	2	ş	39	23	:1		71	46	2::	:	90	1	مند	*	35	50	5- -	:	104	63	***	:
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	,10 ±	2-1	37	2. 30	:			4.0	:	СŲ	42		;	11	50	549	;	36	5,5	est.	:	95	67	MD.	:
		54	36	2.4	1		*	4		1.5	32	*	2	e.	·Pa		*	37	54	-		-	•	~	
				4, 5,45,1,6		. 5	a service of		1		40 G G C 1 7 1 1				enders of the contract	The second second		APPENDING	er - men unan				-		

SHADS TECH BATURE (ZAHRUMTSIT) AND BUASURN INCOP RAINFALL FOR 1938.

		July			ugust		pterb			ctobe			"eນc			cemo		- · ·
Dote:	Kex.	Min.	Rain Fell		Hin.	Rain Fall	hin,	Rain 'all		Min.	Rein Full	: %ax.	lin,	Ruin Yull		.in.	Rein Fell	
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.00 : 1:	105	6 s 5/k		1 33 104	54 65		45	4/5	: 3 1 81	56 54	4 2	64	45	 100	54 57	37 24		:

SHADE TWIETRATURE (PAHE USIT) AS SOUTH IF US

41. TABLE)

MONTHLY WEATHER REPORTS

HALFA - REFINERY SITE DATA SUPPLIED BY C.R.L. HAIFA REPORTED BY W.P. BELL

	1	9 4	æ	1	9 4	3	1	9 4	4
	Tempera	tures ^o F	Humidity	Tempera	tures ^o F	Humidity	Tempera	tureacr	Humidity
	Dry Bulb	Wet Bulb	1 %	Dry Bulb	Wet Bulb	%	Dry Bulb	Wet Bulb	%
January				59	57.5	89	60	5 €	79
February	62	54.5	63	60	58	87	62	55	65
March	65.5	56	58	61	5 9-5	92.5	67	60	68
April	73.	63.5	56	65	64	93	72	(66)	(73)
May	80	69	58	75	68	70	76	(70)	(73)
June	87	74.	5%	84	75	68	. 82	(73)	(72)
July	89	74.	52	85	76	68	86	(79)	(72)
August	87	74	55	88	78	64.5	82	73	66
September	86	75.5	65	87.5	77	63			
October	84	71.5	56	86	74-5	59			
Novembar.	79	69	55	71	68	63			l
December	57	53	80	. 70	64	72			-

Remarks: (1) Figures in brackets are probably too high.

(a) Omitted figures were not registered at the time
(3) This table contitud in Summarised Data on Fami
Absolute or Extreme Values not given.

. 2 ..

Уэсг	Press. @			T s	мре	r e	ı t	u r e	op			Relat	ive Hun	udi ty	Ra	i n r ,	1 1)
d t.	Mean	Average	Dry B	ilb Temp.	Extre	me Dry	Bull	Temp.	Average	Wet B	ılb Temp.		%			Highest in	24 hours
M ov th	(unches)	Day	Night	Mean	Kax.	Data	Min.	Date	Day	Night	Mean	Day	Night	Mean	Motal	Inches	Dute
July August September October November December 1947-January February Marsh April May Juna July August	29,77 29,76 29,85 29,98 30,06 30,05 29,98 30,02 30,04 30,03 49,98 29,98 29,78 29,84	86 87,5 86 80 76,5 59 63 70 74 80 84 87 88	77 78 77.5 69.5 64 59 54 555 563 69 77 77	815 83 82 75 70 61 565 59 645 745 785 825		15:8	54	19:7 9:8 30:9 31:10 14:11 21:12 14:1 10:2 16:3 17:4 15:55 22:6 21:7 8:8	77 78 75.5 68.5 65.5 56.5 57 62 63.5 69 72 76 78	73 74 72.5 63.5 59.5 53.5 51.5 53 55 63 68 73 74	75 76 76 66 62.5 55 55 58.5 61 66 70 74.5	66 66 62 57 57 68 70 66 58 57 61 64	83 83 79 73 78 70 85 90 85 75 72 77 83 85	75 74 70 65 67 69 79 80 75 66 65 72 75	NIL NIL 0.396 trace 4.335 10.556 1.405 0.460 0.603 0.210 NIL NIL	0.190 trace 1.542 2.890 0.463 0.232 0.256 0.116	14:10 29:1i 14:12 13:1 17:2 5:3 16:4 12:5

Remarks: Omitted figures were not registered at the time.

TABLE III

PALESTINE METEOROLOGICAL SERVICE.

MONTHLY WEATHER REPORTS

HAIFA MT. CARMEL

					-Cathorne	SUPPLIE	D BY I	PALES	TIN	E GOVE	RNMIT							
Y-ar	x) Pr⊲ssure			те	M P	E R	А Т	U R	E G	F		Relat	ive]	E um	idity	Rs	infa	1 1
&	Mean	G.	М ,	T .	h	EAN		ß	X T R	E M	E	G.	¥	T		Total	Highest in 24	Hours
Month	Inches	0 6	12	18	Мыхо	Mi.n.	Mean	Mex.	Da+s	Min.	Dat=	06	12	1.8	Mean	Inobes	Inches	Date
1946 December	29.94	55	63	58	64	52	58	72	2:12	46	22:12	69	59	69	66	4.095	1350	14:12
1947 January	29.90	52	56	54	59	50	54.5	68	27:1	44.5	14:1	77	74	75	55	11.949	3.185	13:1
February	29,96	55	61	56	63	50	56.5	75.5	28:2	43	4:2	69	68	76	7.1	1.145	n. 362	17:2
March	29.97	60.5	66	60	69	56	62.5	87	27:3	50.5	16:3	69	60	75	68	0.386	0.200	17:3
April	29,96	65	69	63	72	60	66	92	11:4	49	17:4	62	60	69	64	0.524	0.240	16:4
May	29.86	71.5	75	68.5	79	64.5	72.5	95	4:5	59	13:5	63	57	72	63	0.240	0.086	9:5
June	29,85	73.5	77	71	79.5	68	-/4	84	28:6	64	2:6	784	69	85	76		-	-
			ļ	-		1	}								1	1	J	<u> </u>

x) Corrected to sea level.

60.

(5) KISCLYANSOUS CATHATIC DATA

5a.

61

irecipi tation

Moer Armual Rainfall

	Felght		4 4	A
Station	in foet (approx _e)	OS.	Teriod	Authority
er aus de la company de la	(WONLOW)	TAZIII OL	a fig. 10 i jungstags (mill) dar vig de med edaletakensk dette kolo e ett.	pierus video seminamininto video (s. 1500 s. 1 Tanto seminamininto seminamininto seminamininto seminamininto seminamininto seminamininto seminamininto semina
Lower Belta				
Pao	7	6,6	1936-9	Fort
Magil	7	5.2	1928-39	Mailways
Shualba	60		1926-59 1925-59	Rodallo.
Ghubai shiye	13	~ ~	1928-39	Asilwoys
Ur	2.3	2.9	t!	tr.
Amera	1.5 7.3 30	8.3	1936 -9	osto and Teleg ra pha
Stranger The PAG				
Upper Dolta Opmawa	%0	3.0	2928-39	Railways
	70	4.9	H H	R. A.F.
Dimmiya	9 0	Š.9	15	Reilways
Hills	55 55	2.4.	st	out sin temps or
Kardala	132	3.3	\$ *	Irrigation
Kindiya		5.8		
Rabbanlya	U.L.	7.8 6.0	1937-9	R.A.F. Fosts and Tolographs
Gola Silar	43	32	193 €-9	nquiquetta ana suagrapa:
Sut ol Impra	52 33.0	5.9		79 A 13
Hinaidi (Bagbded)	12.0		:928- 59	
Samarya	22.3	3.3		Railways
Mandali	350	14.4	1936-9	Posts and Telegraph
Janira and Euphrates abo Raredi	7C			
Balji	459	3.5	1.934,-9	InfoGo
Sinjer	1,950	19.5	1956 -9	fosts and Telegraph
Bir Ugla	1,280	17.2	1956-9 1956-8	103100
Eeditha	& 50	5.6	1934-9	L.F.C.
Ana	500	5.4	1936-9	osts and Telegraph
Assyrian Plains and Foot	9407 m			
Table St. (Mansut)	220	6.7	3920 ~39	Reilwoys
Jaraghen (Jalonla)	390	7.3	کور فیل ۱۳۰۰ مرید اداده 19	19
Thanagin (account)	660	75°0	1951-9	+1
Tus Khurmatli	720	8.3	3928-39	66
Iftikhar	670		ا من المنظم المنظم المنظم المنظم	n
TT T 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1,0 0		17	I.P.C. and Railways
Kirkuk			3,936-9	I.E.G.
Ditio	780 7 % 60		1936-8	Fosts and Felegraph
Srbil	1,360		2925-39	Rahara
Mosul.	730	13.0	ニッペンツンツ	tha thi q K u
Kurdish Mountains			عد احداث بريو	
Helabja	2, 300	63.3	1935-9	ibsts and Telograph
Sulalespiya	2,750			8
Diena (Ruwandis)	್ಕ 700	Mich	1936-7,	¥1
			2539	
Agra	2 ,500		1935-9	
ů močila	3,50 0	43.2	()	11
Zakho	I,450	40.4	79	11
Woutern and Southern Des				
T1	3,040	6.6	1034-2	I.P.C.
10.	2,340	5.9		n
112	i,950	5.7	:	et
H3	2,550	4.8	t. t	n
ોપ ાં ઇલ	a, 080	5.8	1929 -39	R.A.F.
Sukhaib	3°000	3.7	1936-9	Folice
ခြားခြင့်သူ့ Approved For Release 20	980		12	:1

Precipitation Rainfall in Inches and Number of Rain-days

	Jan.	reb.	Mar.	Apr.	May	June	July	Aug.	Sept.	Oct.	Nov.	Dec.	Year	Nax. in 24hrs.	
SHUAIBA (13-14, yea Inches Days		servat	ions)		0.1	0.0	0.0	0,0	0.0	0.1	1.1	1.0 3.7	5.7 21.3	2 •2	
DIWANIYA (10-12 ye Inches Days	ars; 0	bser¶s 0.9	tions) 0.2	0.5	0.7	0.0	0.0	0.0	0.0	0.0	0.6	0,9	4.9	4.5 æ	
HINAIDI (13-14 yea Inches Days	1.2 4.9	1.1	iona) 0.3 2.6	0.4 2.7	0.4 1.8			0.0	0.0	0.1	1.0	1.0 4.5	5. 5 28.3	1.5	IKA
RUTEM (9-10 years' Inches Days	obset 0.9 4.3	0.7	0.1 1.0	0.4 3.0	0.2	0.0		0.0	0.0	0.1	0.7 3.7	0.7 3.5	3.8 20.5		B _r
MOSUL (13-14 years Inches Days	obse 2.1 8.8	3.0	1.6	1.8 7.4	0.5 3.1	0.0		0.0	0.0	0.2	1.8 7.8	2.0 9.1	13.0 59.9		

In the rainfall, the figure 0.0 indicates a mean rainfall of less than 0.05; in the rain-days the figure 0.0 indicates a total of less than 5 days in the month over a period of 100 years, 1.e. less than 1 rain-day in 20 years.

Abnormal thunderstorm in May.

Abnormal downpour in November.

5b.
SYRIA & LEBANON.
Precipitation
Rainfall (inches)

The state of the s	Yra.	Jan.	Feb.	Na r .	Apr.	ау	June	July	Aug.	Sept.	0c t .	Nov.	Dec.	Total
Coast Alexandretta Beirut Haifa	10-11 61 14	3.0 7.3 7.1	3.5 6.4 5.7	2.5 3.5 0.9	2.2 2.2 0.7	1.8 0.6 0.1	1.5 0.1 0.0	0.5 0.0 0.0	0.6 0.0 0.0	2 .0 0.2 0.0	2.9 1.9 0.5	2.9 5.1 2.7	3.9 7.5 6.7	27.3 34.8 24.4
Mountains El Kareya	10	11.1	13.7	7.7	3.7	1.4	0.3	0.0	0.0	0 ,3	2.0	6.5	10.0	56.7
Depression Homs Kaara	6=10 8=11	2 .9 5.6	3.3 6.0	1.0	1.0 2.5	0,2 0.3	0.0	0.0	0.0	0. 0 4 0.1	1.0	1.8	2.0 4.5	13.4 24.7
Steppe and D Aleppo Selemiyeh Damascus Palmyra Heir ez Zor Urfe	9sert 5-7 2? 7-10 6-9 5-10	3.0 2.4 1.7 1.0 1.6 2.6	2.8 3.5 2.1 0.8 1.0 5.0	1.6	1.3 1.2 0.5 0.6 1.3	0.4 0.8 0.2 0.3 0.1	0.0 0.0 0.0 0.0	0.0	0.1 0.0 0.0 0.0 0.0	0.0 0.7 0.0 0.0 0.0	0.8 0.4 0.4 0.3 0.2 0.4	2.4 0.5 1.6 0.3 1.9	3.2 2.2 1.6 1.1 1.0 2.9	15.1 12.8 9.2 4.5 6.3 16.1

SYRIA & LEBANCE.

Lean Number of Rein-days x

and the second s	Yra, obana,	ฮัย ม ∙	ಚಿತ್ರ.	mer.	Apr.	lisy	June	لإلمنات	۸ نفق ۸	Sept.	Oat.	liov .	Des.	Total
ramor Si azamirette Batrui (447a	3 61 14	9 15 16	12. 14. 13	8 11 5	7 6 4	5 3	L 1 0	2 1 0	5	5 1 0	6 4	8 96	9 13 11	76 78 56
indistriction Tolknown	10	15	12	31).	7	5	2	0,2	0	ĩ	6	30	12	5k
oepesauton Coma Coma	2-7 8-1)	24	11 12	99	5	0.7	0.1 0.7	Ω Ω	0 .1	0.1	2 3	iı a	9	50 68
Stocke and be Class; Socialized Socialized Paragra Socialized Socialized Socialized Socialized Socialized Socialized	5-7 29 7-8 5-8 5-8 7	11 12 7 6 6 8	11 15 4 5	7823331	453339	2 0.7 0.9 6	0.5 0.1 0.3 0.4 1	0 0 0.3 0.0	0.2	0 0 2 0 0	3223543	745448	11 10 56 59	56 60 33 31 29 65

m Rain-day is one with 0,008 inch rainfall or more.

Musber of Days with Hail

en 1980 in the contract of the	Frs. Sons.	Jan.	e deb	Mar and	jiay_	June	July Aug.	ါတ္မွန္ ေ	Oct.	Nov.	Juc.	lotal
	7 E	1,2	1.6	7 5 O.L	0.1	0,1	0.0 0.0	വുക	0.1	14	1,2	6.7
045年表生38 14 京生時報と 機 。	4			3.6 0.6								
The second secon	**************************************						- 4					

Sc. 17 TVG Precipitation (in inches)

	Jan,	Feb.	lar.	Apr.	May	Juno	July	Aug.	. ept.	Oct.	Nov.	Deo.	Year
1. Coast Acre(1928-38)													
Mean	5.5 1.2	7.9	0.9	0.7	0.2	trace	0.0	0.0	0,2	1.3	4.1	3.7	24.5
Hax, in 24 hrs.	1,2	1.1	0.5	0.4	0.1	trace	0.0	0.0	0,1	0.3	0.9	1.4	
Haifa(1921-34)	7.1	5.7	0.9	0.7	0.1	0.0	0.0	0.0	0.0	0.5	2.7	6.7	24.4
Max. in 24 hrs.	3.4	2.2	ĭ.ŏ	ĭ.o	0.4	0.2	0.3	0.1	0,1	1.9	2,6	7.2	-
m hax, in month	12,1	10.5	6.0	3.5	2.1	0.6	0.0	<0.1	0.5	3.3	13.6	13.9	39 .5
and year													
m Ein.in month	0.7	0,2	0.9	<0.1	0.0	0.0	0.0	0.0	0,0	0.0	< 0.1	0.0	16.8
and year													
Jaffa(1902-12) Hean	5.6	3.9	2.7	0.9	0.1	40.1	0.0	0,0	0.2	1.4	3.2	ئ، ق	23.6
liax. in month	8.6	5.9	7.6	2.0	0.6	0,2	0.0	0.0	1.1	3, 9	5.5	12.5	28.4
and year	0,0	20.7	(40		•••	0,	~ 5			5.5			
& Kin. in month	4.0	1.0	<0.1	<0.1	0.0	0.0	0.0	0.0	0.0	0.0	$0.l_1$	1.9	17.3
and year													
Gaza(1921-34)										A **	- 1		2.5
lean	3.6	3.4	0.7	0.7	0.1	0.0	0.0	0.0	0.0	و.0 ز.1	1.4 2.1	5.6 2.9	3.3 .8
Max. in 24 hrs.	2.7	1.8	0,8	2.7	0.4	0.1	0.0	0,0	0.0	ر ماد	e è a la	۵۰ ۶	_
2. Inland													
Jenin(1921-34)													
Hean	5.1	5.4	1.1	1.2	0.1	0.0	0.0	0.0	0.0	9.3	1.7	4.1	19.0
Max. in 24 hre.	2.2	2.6	1.1	2.5	0.5	0.1	0,0	0.1	0.0	1.1	ت. 3	2.1	571
Nazareth(1891-1907 Mean	e* • • • • • • • • • • • • • • • • • • •	1. C	., .,	3 3	2.2	0.0	0.0	0.0	0, 0	6.0	3.4	7.1	27.1
Mean Max, in month	6.3 34,2	4.6 10 .3	3 .7 5.9	1.0 2.8	0.2	0,0 0,0	< 0.1	0.0	$\langle 0, 1 \rangle$	2,5	9,1	12.8	37.9
and year	مد و چارانه	#100 J	200		Je t	Ust	~ W. d	7.0	€ Mariyata	قر : ۵۰	2.4	44	2.02
in in month and year	1.2	3.7	€0.3	0,0	0.0	e,e	⊙. o	0.0	0.0	0,0	0,3	0.7	18.5

PALOSTRIS
Precipitation (in inches) (Cont'd)

	Jen.	Feb.	lar.	Apr.	Lay	June	July	Aug.	Sept.	Oct.	Nov.	Dec.	Year
Beit Jemal(1930-8)		a game and a 1 to be a game and a 1 or	-	The state of the s	. was write the second	THE THE STATE OF T	*** ** *** **** **** *****	ACOMINE INTERNATION AND SEE	FRITTLE DE BENEFIT STATE				
Mean	4.3	4.0	1.3	0.7	0.1	0.0	0.0	0.0	trace	0.6	2.8	2.5	16.1
Max. in 24 hrs.	1.2	1.5	0.6	0.5	0.1	0.0	0.0	0.0	trace	0.3	ī.4	5.8	7027
Jerusalem(1918-34) Lean	5. 3	E 7	2 2	Vi 42.						_	•		
Max. in 24 hrs.	4.1	5.3	1.7	1.0	0,1	0.0	0,0	0.0	0.0	0.2	1.2	2.9	15.9
Laz in month	3.9 14.5	3.4 12.6	1.1	3.5 3.5	0.5	0,1	0.0	trace	0.4	0.9	2.2	3.0	
and year	140,7	ب ن عبد	4500	G 50	1.3	0.2	0.0	0.1	0.8	2.3	8,0	16.5	41.6
4 Min. in month	0.1	0.2	0.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	40.1	0.5	13.3
and year		•		- • •	•,•	.,,,		0,0	0.0	0,0	~ O	0, 9	13.3
1 Latrum(1901-12)													
Mean	6.5	3.3	2, {	1.3	0.3	40.1	0.0	0.0	<0.1	0.9	2.4	5.2	22,8
Max. in 24 lus. Max. in month	4.6 8.9	2.1	1,€	1.9	0.5	0.0	0.0	0.0	0.3	1.4	1.6	3.1	-
and year	0.9	7.7	6.5	4.0	1.6	0.2	0.0	0.0	0, 1	4.5	7.0	12.5	32.5
Min. in month	3.4	0.0	0.6	0,2	0.0	0.0				•			
and year	2.4	0.0	Oec	0,2	0.0	0.0	0.0	0.0	0.0	0,0	0.0	2.9	15.8
Tebron(1896-1914)													
Heen	. 6,2	4.6	3.4 8.6	3,0	0.3	<0.1	0.0	0.0	< 0.1	0.5	2.1	5.1	24.3
Max. in month	13,9	12.4	მ.€	8.3	2.2	1.3	0.0	0.0	0.2	2.5	6. î	14.1	39.8
and year		-00.								•			2,00
Min.in month and year	1.6	0, 2	0.2	<0.2	0.0	$Q_{\bullet}Q$	0,,0	0.0	0.0	0.0	0.0	1.1	16.3
Beersheba(1921-34)													_
Mean	1.9	2.8	0.7	0.h	0.1	m n	0.0	0.0			6. 0		
Max, in 24 hrs.	2.6	1.	1.(1.0	1.0	0,0	0.0	0.0	0.0	0.1	0,8	1.6	7.0
		474 FF 11	ederic ⊆	55.6 9	ALC U	V ₂ U	UoC	0.0	trace	0.4	1.4	2.5	644

Pales IX.2 Precipitation (in inches) (Cont'd)

	Jan.	Pa b .	llar.	apr.	Hay	June	July	Aug.	čept.	Oct.	Nov.	Dec.	Year
3. Jordan Valley		ar richtend, journal Palls - She	Man Artiffuglandin sunige	***************************************	- Name of State of St								
Tiberias(1890-1907)	4.7	3.2	2,€	1, 1	0.2	0,0	0.0	0,0	(0.1	0.6	2.9	5.1	20,2
# Max. in 24 hrs.	2.7	1.2	2,0	1.5	0.4	0.0	0.0	0.0	(0.1	0.5 2.1	1.6 6.7	2.4 8.8	27.7
Max. in month	11.2	6.5	5.0	3.0	0.9	0.0	0.0	0.0	ζ0.1	C	0, 3	0,0	C101
and year Min, in month	0.5	<0.1	0.1	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.7	14.4
and year	- 0,5	, - •											
Beisan(1930-8)	7 7	2,8	0.7	0,4	0.2	0.0	0.0	0.0	0.0	0.7	1,2	1.4	10.5
Mean Max.in 24 hre.	3.1 0.9	0.9	o.l.	õ.3	0.1	0.0	0.0	0.0	trace	0.5	0.4	0.6	•
Jericho(1921-34)	• >	7 7	0.4	0.4	0.1	0.0	0.0	0.0	0.0	0.1	0.4	1.2	5.0
Max. in 24 hrs.	1.3 1.0	1.3 1.3	0.4	0.7	1.0	0.3	0.1	0.0	trace	0,2	1.0	1.6	
Dead Sea, North end (1954-7)						0.0	0.0	0.0	O.lu	(0.1	0.4	1.9
hean	0.6	0.4	< 0.3	<0.1	<0.1	0.0	೦,0	0.0	U.U	U a LÇ	~ O	0.4	240 27
Dead Sea, outh end(1935-6	0.3	0.1	0.0	0,0	0.0	0,0	0.0	0.0	0.0	0.3	0.5	$O_a I_{\downarrow}$	1.2
4. Transjordan											•		
Apman(1924-41)		_		0.00	. 7	0.0	0.0	0.0	0. 0	0.3	1.3	1.6	10.5
/ i san	2.5	3. 3 2. 3	0, { 1, L	0.0 1.€	J.1 0.9	0,0	0,0 0,0	0.0	0,0	0.4	5.1	2.1	~ ·
Fax in 24 hrs.	esy. ∞ marmorn	ere meneralemente suc. Gr. O. (1)	nder frankliche von der	green own stand in sort it.	. ::33/1007	or and the management	management of parties and the sea	and the second s	WIRE AN ARTHUR AND ART	- vacatic minte			Address of the Asset

1334-1505 \$ 1507-15 \$ 1507-15 \$ 1500-95 \$ 1500-12 \$ 2690-9 # 15 years' obsus. (6) GROUND THE PHRATURES AT ALTAND, IRAO

SUPPLIED BY MIDDLE EAST PIPELINES, LONDON - D. BUCKHAM.

PERIOD DECEMBER 1946 TO HOVELDER 1947.

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

MANAGO RAPUSING CROUND THE PERAPORE AT A DESPET OF A PERF.

	90. 3200 0400 2600 0800 1000 1200 7400 1600 1860 2000 22.00 Brs.													
i ing	Gr ₂	0200	0440	0 6 00	9080	10 00	1200	7 4. 00	1600	1860	2000	22.00	Hrs.	
. gs.		76.8	76.7	76.7	77-9	78.1	78.6	78.8	78-8	76.3	77.7	77⊲5	O.S.	
, T.	ۏڒ	5 5	55	55	57	58	59	60	58	58	59	3 5	O ys	
	90	25	93	93	96	97	97	96	96	95		95	Opt	
	Lahur rayan sebesa			Comments of the Comments of th		1		Lancardon Pelatronicion	Startinger State (2) - editing) Ingir (grienisk inks 1821)	Lucian	and remember 1735		

Yearly Characteristics.

Average 77.7°F. Ninimum 55 °F. Faxinum 97 °F.

The Sigure of 50°F for the minimum temperature is not sensitived subject to the sensitive at the line. For this reason it seems probable that the singurant sigure about be taken as 66°F which was registered on Settleman 1967. An error in the instrument has been confirmed from side. [Aboden memo 37075 of 26th January, 1948.]

ALWAND REPINERY SHADE TE FERATURES.

4.89	Olio			1	1		}	24,00	•	3600		-11-4-19-19-19-19-19-19-19-19-19-19-19-19-19-	, 14s ta
Will Co	69.2			63.7							79	72.5	o a
, 1029. ,	25	Ju ₊₊ 2	32.5	315	30.5	39	l.ls	4.5	1 19	(a)			£ £gye
sac	102	100	99	98	102	109	117	117	3.1.?	736		i sije	

Monthly Characteristics.

Average 74.9 Minimum 30.5 Maximum 117

Approved For Release 2003/01/29 : CIA-RDP80-00926A005400050022-5

SHEIGHED BY HIPDUR CAST PIPELINES LOW SCHOOL OF BUCKELS

STATISTICAL PESULTS

decourature at 2 hourly intervals.

MANAGE EVELLET GROUND THE PERATURE AT A DETTIL OF A TREET.

ino	oc.	o200	0400	¢600	0600	1000	1200	1400	1600	1500	1000	22.20	75 S
vge.	63.7	63.3	63.3	63 .3	62, 1	619	65.4	6 5 . ?	45 A	9. j.	54,	4,5 £	€gg _
i. y a.,	35	55	55	\$5	57	58	59	50	£8	570	F8	: : Is	7
1980.	69	69	69	69	70	72.	72	71	'n	7u	<i>7</i> 0	70	≎ F i

Daily Temperature Ronge

Monthly Charmatantation

inime 2.96°p inime 1 op printe 5

Jago del

一种种.

Average 65.29F Minimum 75 °F (nonember a 62 g syste) Maximum 72 °F (12N) bes first April)

ALMAND REFINERY - SHADE TELLERATURE

20 mag - 1- 5 asses	MARKENICAN LARGE MARK	water and the second	ments or company or manager										
'trea	00.	0200	ortoo	0600		i		:	t	•			Ey a,
∨go.	61.1	59.4	57.1	56.1	:	1	1	1	:			,	G _g ,
LZ.	51	50	49	47	51	56	59	3 3	-	197	<i>9</i> 7	jyla	*
ax.	76	74	74	72	80	84	86	57	ft5	(3.4)	777.	75	calla
Mattheway regions to the same	C. American de la company	l Orres e communicación de la co	L!	f., l	•	•	•		:	•	,		

Monthly Characteristics.

Average 64.9°F Minimum 47 °F Maximum 89 °F

71.

OCTOBER 1947

S. -41 S. Call Results

Temperature st 2 Hourly Intervals.

ALTER MED AT THAT IN PERSONNE AT A DEPTH OF & PERSON

A North Control		e Sellenterates a series	THE PERSON NAMED IN		region in the same and								
P6	oo.	0200	نه کیلهای	Cécu	£500	1,000	1290	1400	1600	1800	2000	22.00	i ma
·30.	73△5	73.3	73 0	73.6	7%.3.	75.2	75.8	76.0	75.9	74.5	The	13.1	0.5°
	69												
Lo	70				80	80	80	80	So	76	78	 75	
	the second	w	* = *				*. *. <u>.</u> . *	Property of the State of the St	: 	Separate mass of the con-	i	:	

Daily Temperature Rangely.

Average 3.5200 Himmon 8 Haarman 5 % Monthly Characteristics.

Average 7%-307
Minumum 68.007 (0600 mms
16 (067)
Mondamum 80 97 (2000 hora
2 (065)

AL PLANTED AND THE ARTERS

1	1	-				· Hammer way go can	nen salatan ya	Cartella de la constitución de l	ener iv	Themselves	A fire attenue annotation part.	Service Servic	4
NJ .	0 0,	0200	Sec. 1	1600	City of	1,000	1200	1400	1600	1800	2000	22.00	Era.
		Jane 1 m. S. San C.	4	me		en e la constanta	de trace a la compa	44.00		`	1		
71 1	739	i (filmi) L	, 477, 6 k	166 7	72.9	. 93, 2	59.2	20-3	19.6	£25- S	79.6	74. 42	
	75	54		J		5C	76	1.72	74	56	62	ا موارسا	ļa.
	£ .	35¢.		₹.	70	<u> </u>	104	100		95	24	39	2. j.

anddy Characteristics

Creaning 18.2 Colonia G (0600 hrs 26 0cf.) Colonia Chi (Red are cont)

72.

SEPTEMBER 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALVAID REPLIERY GROUND THE PERATURE AT A DEPTH OF 4 FRET.

line	00،	0200	0400	0600	0800	1000	1200	14.00	1600	1800	2000	2 200	Hrs.
/aBe*	83.8	83.7	83.4	1		8 6 ₅ 2		Ì		85.7		84.1	၀ာစ္
in.	77	77	77	76	78	76	79	80	80	78	78	77	ാഷം ാഷ്ട
ar.	88	88	88	87	90	90	90	91	51 .	90	90	8 8	23 €

billy Temperature Range.

Monthly Characteristics.

Average 3.60°F Minimum 2 °F Maximum 5

Average 34,90P Minimum 76 °F (0600 hrs 23 Sept.)
Maximum 91 °F (1400 hrs 1 Sept.)

ALVAND REFLUERY SHADE TELFERATURE

ire	00.	0200	0400	0600	0800	3000	1200	1400	1600		2000	22,50	lirs.
'age.	79.0	76.1	74 ₀ 2	72.0	82.9	91.8	97.2	99.2	98.7	94-9	8.38	83.8	၁ဏ္ဍ
iin.	74	72	68	64.	75	87	90	92	93	86	80	76	oh
'ax.	95	93	90	86	93.	3.06	209	123	1322	104	98	96	Ole

Monthly Characteristics.

Average 86.5 °F
Minimum 64°F (0600 hrs 8 Sept.)
Maximum 312°F(1600 hrs 13 Sept.)

19/31 AUGUST 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWARD REFINERY GROUND THE DERATURE AT A DEPTH OF & FRET.

1399	60s	0500	0400	0600	08 00	1000	1200	27400	Į.	1			Hr e.
*ge.	90.5	90-1	90 .0	90.0	93.3	93.6	9 i i0	94.3		Ř	92.3	•	og:
in.	88	88	87	87	90	90	90	91	91.	30	89	89	e.B.
Ax.	95	93	93	93	95	95	95	95	95	95	95	94	o _k

Daily Temperature Range.

Konthly Characteristics.

Average 4.30F Minimum 2 of Meximum 7 of Average 92.207 Minimum 87 97 (06.00 km 30 Aug.) Marinum 95 97 (06.00 km 19 Aug.)

ALWAYD REFLERY SHADE TELLERATURE

LEDIO .	co.	0200	04.00	0600	0800	1000	1200	1400	1600	1800	į	22.00	į.
₹ge.	86,8	83.2	80.1	77.7	91.4	i i	104.9		1	1		93.3	o _j
in	80	76	72	70	82	92	101	103	103	100	92	87	O _f
AT.	96	94		93	101	108	112	114	113	111	104	99	ဝဋ္ဌာ

Monthly Characteristics.

Average %, 407 Minimum 70 °F (0600 hrs 51 Ang.) Maximum 114 °F (1400 hrs 26 Aug.)

JULY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REPTREMY GROUND TESTPERATURE AT A DEPTH OF 4 FERT.

2 Face	: 90 ₅	0200	04,00	0600	0800	2.000	1200	1400	16 00	1800	2000	22.00	Brac
.28a -	91.0	50.6	90.5	90.8	93.7	94-0	94.0	94.1	94-3	94.0	92.6	91.9	ပုံ
ja.n.	85	. 35	36	\$7	92	92	92	92	92	92	20	9 ে	o _y .
3).T	Ör.	95	93	93	%	97	97	96	96	9 5	95	95	O _E

Dy Temperature Range.

Monthly Characteristics.

Average 3.56°F Minimum 3 °F Caminum 5 °F Average 92.7°F Minimum 85 °F (0000 hrs 31 July) Maximum 97 °F (1000 hrs 25 July)

ALMAND REFINERY SHADE TENPERATURE

inve	00-	0200	0400	36 00	9800	1000	1200	1400	1600	1800	2000	2200	Dra.
je.	98.4	67°.	62.1	83.6	95.0	102.6	108.9	110.6	110.4	108.5	106.4	97.a.	្សា ខ្សា
a.	85	80	73	77	9 0	9 6	104	105	105	LO4	55	92	ஷ
4 0	102	100	99	98	102	109	127	£17	117	I)h	107	101	بيرن

Monthly Characteristics.

Average 98.8°F

Sinimm 73 °F (0400 hes 4 July)

Earlaum 117 °F (1200 hes 20 July)

JUNE 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWARD REPRESENT GROUND THE PERATURE AT A DEPTH OF 4 PEET.

june	20.	0200	0400	0 600	0800	1000	1200	1400	1600		2000	22,00	Hra.
age.	85°7	85.3	85 。0	85°6	87.5	87.9	89.1	88.5	80.5			8).9·	Оую
170.	87	86	86	87	89	89	89	90	90	88	87	87	$\alpha_{\mathbb{P}}$
83%	92	9ïT	90	31	93	94	94	94	95	94.	92	92	ပန္တာ

Diurnal Temperature Range.

Monthly Characteristics

Daily 3.5807 Minimum 3 Op Naximum 5 Of Average 86.8°F Minimum. 36 °F (OMGO into 17 June) Maximum 95 °F (1600 into 16 June)

ALMAND REPTUERT SHADE TELEBRATURE

ERO La collection	00.	C 200	0400	c600	0800	1000	3.200	11,00	2,600	1800	5 000	120 M	in a
ൂളം	85.6	82.2	79.4	79.6	93.4	99.7	1.02.5	1046					Бра
$\mathfrak{M}_{\mathfrak{o}}$	80	77	70	70	86	88	30	93	97	94.	38	82	oħ.
-sur _c	91	88	87	90	100	106	110	114	2.14	130	104	97	i ⊅fe i

Monthly Characteristics.

Average 93.40F
Minimum 70 OF (0400 hrs 5 June)
Maximum 124 OF (1400 hrs 28 June)

MAY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALMAND REFLICTOR GROUND TEMPERATURE AT A DEPTH OF 4 FEET.

ise	00.	02 00	04.00	0600		1000				1800	2000	22.00	Hrs.
rge.	83.9	83.1	83.1	35,0	83.3	83.8	849	85.0	8k-9	846	83.8	83.4	Oge
in.	79.5	79	73	74.5	77.0	79.0	7 9 .5	79 _e 5	<i>7</i> 9∘5	79.5	79.5	79.5	OF
AE.	88	88	83	88	88	88	90	90	90	90	88	88	Op

Daily Temperature Range.

Monthly Characteristics.

Average 2.10p Minimum 1.00F Maximum 7.5°F

Average 85.9 of Minimum 73 Maximus

ALUAND REFINERY SHADE TESPERATURE

Etra.	:	1											
	22,00	2000	1800	1600	14.00	1200	1000	0800	c 600	0400	0200	00,	imo
or	83.6	87.9	93.1	94.7	94.9	92.9	89.8	84.0	72.7	73.5	7 5.7	79.5	780.
ck	67.5	69	74	74	72	68	62.5	62	56	59	60	ഖ	in.
c.Ma	98	98	103	104	104	105	304	94	90	84	87	92	ax.
	67.5	69	74	74	72	68 105	62.5 104	62 94	56 90	59 84	60 87	ຄ	in.

Monthly Characteristics.

Average 85.17F
Minimum 56 °P (0600 hrs 16 May)
Maximum 105 °F (1200 hrs 30 May)

AFRIL 1947.

STATISTICAL RESULTS

Comparature at 2 Hourly Intervals.

PLEASE DEFINERY GROUND TO PERATURE AT A DEFIN OF & FEET.

	ж. Эё.	02 00	2400	osce	0800	1000	1200	1400	1600	1800	2000	2 2.00	Hrs.
್ಷಾರೆ.	16.2	76.1	76.3	76.5	76.5	76.6	76.7	76.7	76.E	76.3	76.0	75 ₃8	op
k. ž.	/2	G.	71	72	70	70.5	71.5	71.5	71.5	71.5	71.5	72	oğ.
#	79	79	79.5	79.5	79	30	80	80	79.5	79.5	79	79	oge

on an Temperation Range

Monthly Characteristics.

warage 1.90g 3.50g 3.50g Average 76.39F Minimum 70 9F (0800 bre 1 Apl) Maximum 80 9F (1200 bre 23 Apl)

AGRAND REFINERY SHADE TEREBRATURE

77	00.	020 0	U+00	0600	0800	1000	3.200	:ULDC	2,600	1800	2000	22, 00	II:50
#uztu.	55.3	5 <u>L</u> 1	61,6	62,1	73-7	31.3	85.1	86.1	8 5.0	82.8	76_3	70.5	OT
\$ F.	54	5 2	50	5C	58	64	69.5	70.	71	70	65	60	OF
2.5	36	83	86	84	85	977	985	39	96	94	92	86.5	ep

Monthly Characteristics.

Avonago 74. 60%

Marriage 90 95 (Oa00 hrs 18 Apl) Mariage 99 95 (1400 hrs 26 Apl)

MARCH 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFINERY GROUND TREFERATURE AS A DESTRICT A MEET

Time	00.	0800	0400	0600	0800	1000	1200	1400	1600	1800	2000	22.00	Has
Avge.	59.0	6 8 .8	68.7	68,6	69 .0	69, 2	69.3	696	69.4	693	69.1	69. C	οp
Min.	ő 6, 5	66	66	66	66.5	66.5	67	67	67.5	67.5	67	67	्राह्
Maz.	72	72	71.5	71.5	71	72	72	72	71.5	72.5	71.5	72.5	Oja

Unily Temperature Range.

Monthly Characteristics.

Average 1.807 Minimum 0.507 Maximum 5.007

Average 600% Minimum 66°F (0200 kms 1 March) Maximum 72°F (1000 kms 19 March)

ALMAND REFINERY SHADE TENTERATURE

Time	∞,	0200	O1*00	0600	0300	3000	1200	1400	3600		2000		Ħs
Arge	59.3	57.8	56.4	55.1	50.5	68. 0	77.4	73,0	72 3	692	63.0	62.5	Om
Nin	5 2	50	&B	46	50	55	50-5	שנ	58	58	57	55	Ope
Max	72	73.5	74	74	70	86	86	36	35	84	76	273	0 я

Monthly Characteristics.

Awarage 63.9°F Himmum 66 °F (0600 bris 2 March) Maximum 86 °F (1200 bris 22 March)

FRERUARY 1947.

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFINERY GROUND TEMPERATURE AT A DEPTH OF & MEET

Pline	00.	0200	0400	060c	0800	7000	1500	1400	2600	1800	8060	22.00	Ars.
æ.	66.7	66.5	66.5	65,5	66., 6	66.8	67.0	67.2	67.4	67.4	67.0	66.9	Ob
ir.	66 .	66	66	66	66	66	665	66.5	66.5	66	S 6	66	Og
3 3.	68	67.5	67.5	67.5	67.5	1		}	68∘5	l 1		68.5	Ope

Daily Temperature Range.

Monthly Characteristics.

Minimum .5°F Kazimum 2.0°F Average 66.60F Minimum 66 °F (0000 hrs 10 Feb) Maximum 68.5°F (1600 hrs 2 Feb)

ALMAND SHADE TERESTATURE

ime	cc.	0200	04.00	0600	6800	1000	1200	2400	1600		2000	22.60	ne.
wge.	49.2	47.6	46.3	44.6	47.6	53.4	58,8	61.8	61.8			51.7	OF.
ilin.	39	36	3h	32.5	34	40	44	45	47	43	لجل	40	Gg-
ex.	60	57	57	55	55	63	69.5	72.5	74.	70	63	52.5	og.

Monthly Characteristics

Average 48.80P

Minimum 32.5°F (0600 hrs 5 Feb) Maximum 74 °F (1600 hrs 15 Feb)

JANUARY 1947

STATISTICAL RESULTS

Temperature et 2 Hourly Intervals.

LIFAUD REFUSERY GROUND THE PATURE AT A DEPTH OF L FEET

2,010	00.	0200	0400	0600	08 00	1000	3.200	1400	1600	1800	2000	22,00	Hra
i sges	69. D	53.9	68.8	68 _e 6	68.4	68.7	68.9	69,2	69.5	69.4	69.3	69.2	op
Linia	67	57	67	67	67	ક્દ	57.5	68	6 8	68	67.5	67	Oğı
Nux.	72	71.5	70.5	70	70	71	72.5	71	72	72	72	72.5	op

Daily Temperature Range.

Monthly Characteristics.

Average 2.00F Hardware 2.00F Hardware 5.00F Average 68.9°F Nimiras 67 °F (0800 hrs 4 Jan.) Naximum 72 °F (1600 hrs 4 Jan.)

ALGAED REFLIERT SHADE TEMPERATURE

							****************	and the second	MT. STOWNERS & LARGE	-1			
1 250	C C.	0200	04.00	06 00	03 00	2000	1200	2800		1800	'		Hrs
ge.	51.7	51.1	50.1	50.2	51.1	55.1	588	60.6					OF
: 	43	43	40	36	36	43	50	53	52	49	!	44	OF.
* <u></u>	50	: 39, 5	50	62	66	Gi,	63	69	58	64	66	60	Oğ.

Nonthly Characteristics

Armage 54 tor Ministra 36 °F (0800 hrs 26 Jan.) Maximia 69 °F (1800 hrs 13 Jan.)

17/31 DECEMBER 1946

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ALWAND REFERENCE GROUND THE FERATURE AT A DELTH OF 4 FRET.

00 6	0206	04 00	0600	0800	1000	1200	1400			2000	22,00	Hrs.
72.9	73.0	72.8	72°6	72.5	72.5	72.5	7 5-0	73°4	73.6	73.A	73.2	OF.
71	70.5	70.5	70.0	70	69.8	69	69	71	72.5	72.5	71.0	्रीह
76	7 7	76.5	76	75	75	75	76	76	77	77	77	оÞ
	72.9 71	72.9 73.0 71 70.5	72.9 73.0 72.8 71 70.5 70.5	72.9 73.0 72.8 72.6 71 70.5 70.5 70.0	72.9 73.0 72.8 72.6 72.5 71 70.5 70.5 70.0 70	72.9 73.0 72.8 72.6 72.5 72.5 71 70.5 70.5 70.0 70 69.8	72.9 73.0 72.8 72.6 72.5 72.5 72.5 71 70.5 70.5 70.0 70 69.8 69	72.9 73.0 72.8 72.6 72.5 72.5 72.5 75.0 71 70.5 70.5 70.0 70 69.8 69 69	72.9 73.0 72.8 72.6 72.5 72.5 72.5 75.0 73.4 71 70.5 70.5 70.0 70 69.8 69 69 73.	72.9 73.0 72.8 72.6 72.5 72.5 72.5 73.0 73.4 73.6 71 70.5 70.5 70.0 70 69.8 69 69 72 72.5	72.9 75.0 72.8 72.6 72.5 72.5 72.5 75.0 73.4 73.6 73.4 71. 70.5 70.5 70.0 70 69.8 69 69 72 72.5 72.5 72.5	72.9 73.0 72.8 72.6 72.5 72.5 72.5 73.0 73.4 73.6 73.4 73.2 71. 70.5 70.5 70.0 70 69.8 69 69 72 71.5 71.5 71.0

Diurnal Temperature Range.

Monthly Characteristics.

Average 2.20 Minimum 1.00 P Maximum 4.00 F Avorage 72,90F Minimum 59 °F (1200 kms 31 Dec.) Maximum 77 °F (0200 kms 18 Dec.)

ALMAND REPINERY CHADE THE TERATURE

							Maria Maria Maria (Maria (Maria)	ye disala diken siste say		AND TAXABLE FOR SHAPE	PROGRAMMENT THE TAXABLE	-	,
me	00 _°	0200	Ob.OO	0 600	0800	1.000	1200	14,00	1600	1800		22,00	Ars.
ge.	48.0	l	45 _° 1	44.6	44.6	51.2	58.3	60.8	!			48.9	op
30.	35.0	34.2	32 _° 5	31.5	3 0.6	39.0	49.0	53.0	50.0	47.2	42.0	40 .0	o _F
X ₀	59.0	59.0	58.0	57.8	58.0	64 ,0	740 O	71.c	70°C	62.0	59.0	58.0	oge
	<u> </u>	-	1		Harrier and the same		L	and the second	I williams make		Land residences with	American conserva	

Monthly Characteristics.

American 50.6°F (0800 hrs 25 Dec.)

Maximum 74.0°F (1200 hrs 30 Dec.)

(7) OTH PERFERACULOS AT ABADAN, TRAN

Village to the

intach (***)

TE DE DE PERSON DE PAST PERSONALE DED. LONDON - D. BUCKHAP

NOVEMBER 1947

STATISTICAL RESULTS

Temperature at 2 Hours intervals.

ABADAN DICCHING GRUDY THE MERATURES

(SURVACE LINE SESTER)

****	63.	5800°	Ç#SO	36 05	0000	1000	1200	ucc	1600	1.800	2000	22,00	Hrs.
1774"	57 a	ء جر	50	56 5	58.2	69	80.7	82 · 4	34,	76.6	79.1	65.4	οĕ
1.1	, *5 ,*57	4.7	46	1 ₁₋₂ -	ω ',	52	53	59	62	56	56	52	o _f
	· ·		- **.	73	7.5	. 95	112	136	104	100	89	83	C.Jr

raily Temperature Banes.

Mornidy Cheracteristics.

in opinger (31, 10g) 24 Astron (5, 18) Mandagen (56, 18)

Average 63.609 Birleum 45 97 Daxonn 116 98

A Chartenan The Committees

	in a	0205	02.00	6600	0800	2000	1204	lik00	1500	1800	2000	22.00	Hrs.
	5 7. .5	65.7	િંદુ હૈ	6 2,8	64. ひ	72.5	76.5	90.2	BO.A	75.2	71.5	69	©gy.
	53					1							o _p
ax	Pa.5	80	30	80	Free.	88	93	95	45	87	83.5	88	Ope.

The property of the second of the control of the co

emonege in in in

OCTOBER 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAN INCOLLIG CRUDE TEL TERATURES

(SURFACE LINE SYSTEM)

line	00.	0200	0400		98 09		1200	3 40 0	1600			2 2.00	
ಇಇಂ.	77.7	73.1					102.5	111.4	112.8	103.8	92.0	83.7	oF
An.	67	62	58	56	58	59	74	88	100	90	80	72	o _F
ax.	86	82	76	73	83	306	115	125	128	126	210	96	$c_{\tilde{E}_1}$

Daily Temperature Range.

Monthly Characteristics.

Average 46.19F
Minimum 34 °F October 24th
Maximum 57 °F October 3rd

Average 87.4°F Minamum 56 °F Maximum 128 °F

ABADAN SHADE TELFERATURES

ine	00.	0200	04 00	0600	08 00	1.000	1200	14,00	2600	1200	2000	21.00	Hes.
vge.	77.3	75.5	74.3	72.2	76.0	89.2	97.8	100,6	95.4	33.9	845	80. a	1 4ga
in.	69	66	64.5	63	66	76	87,	3 0	1 107	30	76	72	₽p.
ex.	85	85	82	31.5	38	1.02	1.09	112	112	105	94.	50	gillinga N

Morthly Margatary ettage

Average 84,80p

Minimum 63 % Marriago 212 % (Laro ince a cons)

SKITTEDER 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervals.

ABADAN INCOMING CRUDE THEMERATURES

(SURPACE LIME SYSTEM)

Timo	62 76	O ttoC	රුද්ගර	റദ്രാ	£000	1.200	12:00	1700	1800		2200	2/55 00	Hrs
Avge.	78.9	76.0	73.0	73.7	98.3	133.6	120.6	1200%	110.2	97,6	89.0	83.2	Oφ
Min.	74	71	68	71	90	107	115	111	700	9 0	81	80	ok
llax.	84:	82	82	87	108	123	157	131	120	206	97	88	OF.

Daily Temperature Range.

Monthly Characteristics.

Average 47.6°F 39 % 54 % Minimum Maximum

Average 94.95°P Minimum 68 °F Minimum 68 Noximum 131 of

ABADAN SHADE TELEBRATURES

Pinc	co.	0200	0 4.0 0	0600	0800	;	1200	1400	1660	1800		22,00	Are.
Avge.	85.7	83.1	807	78.3	82,8	1				100.4	92.5	87.5	ōy,
Min.	7E	76	73-5	71	74	89	97	99	99	94	86	82	op
Xax.	.93	90	89	86	1	-			115	108	97	94.	

Monthly Characteristics

Average 91.7°F Minimum 71 °F (0600 hrs 21 Sept.) Maximum 115 °F (1400 hrs 13 Sept.)

\$25.52 38aabah \$1.2008 81 490 0000 \$2 28 markadis \$1.2008 62 400 0000 994 511 markadis

Maide Character Character

చేం	s tot	ξOτ	ETT	5°277	Zπ	977	ott	ror	76	96	86	60	· · · · · ·
Ā.,	S°OS	76	86	5-601	. SOT	907	45	99	- 29	*0	S.,	35	
ďs	8° 5 6	8°66	4°LOT	ट रहाः	375.6	770°1°	E°EOT	5.56	97.6	£.8	\$550 	E 75	FG ^E :
onaji.	.53°00	2002	0081	7600	ontre	9971	0001	0081	1090	Onder.	t skippede	To the second	t gg

SINDAR HART BOARD HACAEL

Average 2.02.60p Mathema 75 vg (0600 hrs & Aug.) Mathema 75 vg (1600 hrs 29 Aug.) could system ev et annings ge et annings

Monthly Characterization

the live Membersethere Mange

	ranama nag		AND THE PERSON STATE OF		-				e.comment.wire		64 1		
ಷೆಂ	25											£6	
₫.o.	ቱፀ											- 48	
Ão	0° z 6	o ଃ	०% ०४	4 DET	9'827	\$ '8∄T.	7277 5	5"50"	Z 228	эте	, 28	2724	
.a .1 H	St.,00	0050	0007	0087	OO!"E	oc'/2	750 0	600 k	onan	0000	ocht.	0670	: : 372 :

(PERSUS HOLD HOVERNS)

SEMPLEVALARED BECOME THOUSEN

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STATISTICAL PREDITES

Lygi asodov

CONTESTICAL RESULTS

JULY 19-7

Temperature at 2 Hourly intervals.

ABADAN INCREDE CRUDS IN IRATURES

(SUPPLACE LIFE SYSTEM)

*	•	0260	0400	0600	0800	2000	1200	14 00	2.600	1800	2000	2200	2400	₹ ₹:::28 -
,3		9 3.2	37.3	8 600	9 0,6	200.5	124.5	132. 8	13400	226,5	332.0	102.6	5 6 .5	o p
		9.	33	78	33	27	315	125	127	126	93 .	96	92	ဝင္ခ
٠.	:	က်ဦး	97	93	59	323.	1.50	142	14.7	141	1.28	116	701	Outs.

Verily Summerchane Range

Monthly Characteristics

Averago 50,9°F Mistemen 39 °C Vacassen 63 °C

Average 107 50F Minimum 78 9F Maximum 147 9F

ABADAN SHADE TEMPERATURE

13 (0000	02 90	CHOO	୦6୦	0800	1000	1200	1400	3£00	1000	2-0 00	2800	Mr is
3.	فيلا	92.2	50,1	8 8.5	25-5	1.06.0	112.7	115.2	11 4 09	110. h	101.7	97.2	(SÁ)
ني : ا	89.5	84.	8¥-	82	91	96	107	333	170	200	98	92	-6 ₃₅ -
į,	200	98	95	94	1.04.	116	119	123	124	139	1.09	162	ञ्जू

Correct Characteristics

Transport 124 (1500 bes 3 delp)

"Transport 124 (1500 bes 3 delp)

STATISTICAL RESULTS

JUNE 1947

Temperature at ? Hourly Intervals

ABADAN AMORITHG CRUDE TEMPERATURES

(SURPACE LINE SESTEM)

ire	0200	040 0	0600	0800	1000		1400	Į.	1800	2000	2200	2400	Hat.
vze.	86.5	82.8					1.50-4	1		109.8			0 %
in.	62	78	75	76	94	109	118	129	213	301	91	86	o _{li}
ar.	95	89	84	39	115	130	142	145	143	131	134	105	ंक्ष

Diurnal Temperature Rango

Monthly Characteristics

Average 52.30F Minimum

Average 104,4°F Minimum 75 °F (0600 Hrs. 6 June) Maximum 145 °F (1600 Hrs. 28 June)

ABADAN SHADE TEMPERATURES

	0000	0000	01.00	2000	and the speed		7000	1	2606			***********	_
inc	0000	0200	OLCO	060 0	9 800	1000	1200	1400	2.600	7800	2000	2200	Hys:
vge.	911	89.1	36.4	34.8	Dec 9	106.2	111.2	112.6	111.9	207.8	98.4	93.9	Þjá
in.	86	ଟ୍ୟ	79	76	87	96	99	3.00	100	98	92	86.5	Op.
3 3 %	97	94	92	92			119	120	121.5	117	105	99	r p

Monthly Characteristics

Average 99.0°F Minima 76.0°F (0600 hrs 6 June) Maximum 121.5°F (1600 hrs 28 June)

MAY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervols.

ABADAN INCOMING CRUDS TRAFFRATURES

(SURFACE LINE SYSTEM)

	, cape	ONGO	0600	C80C	10 00	1200	1400	1600	3.800	2000	2200	2400	Ar s
· · · · · · · · · · · · · · · · · · ·	82.2	76,1	75.3	76 .3	90-9	108.2	119.9	124,2	120.7	108.7	%.1	85. 4.	هری
1	60	65	62°	63	72	90	101	101	200	9 0	84	75	op
est.	96	90	:	Í	104	121	134	139	137	126	110	101	عاد

or of Mangazaltura Rango

Monthly Characteristics

Average 104.3°F Minimum 62 °F (0600 hrs 1st May) Baximum 139 °F (1600 hrs 27th May)

ABADAM STADE TEMPERATURE

	0200	,	1	:	1 .	1200	3400	3.600	3.800	2000	2200	Hrs.
. 57.2	81. 3	79.5	78.2	87.2	96.3	101 .3	1		99,2	92.4	86.6	O _{ff}
		!				85 114			82 113	* *	75 97	ीक्षा ंपूर

Martidy Characteristics

Average 90.898
Manumum 66 9 (0600 hrs 16 Way)
An area are for 11400 ner as may)

APRIL 1947

STATISTICAL RESULTS

Temperature at 2 Hourly Intervels.

ABADAN INCOMING CRUDE TEMPERATURES

(SURFACE LINE SYSTEM)

	M.	0200	04:00	0600	0800	1000	1200	1400	1	1800	2000	2200	·	Hrs.
•	75.6	75.7	69 ₃ 4 ₄	66.4	67.2	77.1	97.0	103.6				85.7		į
	6 2	58	56	5 5	55	64.	78	92	93	87	85	72	68	Op
	85	87	78	75	105	106	115	123	127	124.	110	99	90	o _F

durnal Temperature Range

Monthly Characteristics

Average 47.10F Mindman 28 OF Morimum 60 OF

Average 85.7°F Minimum 55 °F (0600 hrs 19 April) Meximum 127 °F (1600 hrs 25 April)

BADAN SHADE TEMPERATURES

lno	0000	0200	04 00	0600	0800	1000	1200	1400	1600	1800	2000	2200	Ers.
7ge.	75.1	72.0	69,9	68.0	74.8	85∘&	91.h	93.6	93.4	89.8	82.8	78.8	Op
in.	64.	62	61.5	57	66	75	77	77	76	74	70	67	o.B
az,	85	83	80	79	83	97	102,5	103.5	104	98	93	68	ிழ

Monthly Characteristics

Average 81.25°F
Minimum 57 °F (0600 hrs 23 April)
Marimum 104 °F (1600 hrs 27 April)

STATISTICAL RESULTS

MARCH 1947

Temperature at 2 Hourly intervals.

ABADAT INCOMING CRUDE TEMPERATURES

(SURFACE LINE SYSTEM)

line	6000	0200	0400	0600	0300	1000	1200	1400	1600	1800	2000	2200	Hrs.
iin.	(62.0 56.0 70	T I	5%.1 51 65	63.4 49 80	77.7 57 9 7	90.0 70 108	95.6 65 109	91.2 66 104	86.8 67 96	76.9 65 88	70. 5 63 76	op op

Divinal Temperatures Range

Monthly Characteristics

Average 39.10F Mintmum 5 Maximum 48 O.P Average 74.70F Minimum 49 °F (0800 hrs 2 March) Maximum 109 °F (1400 hrs 27 March)

ABADAN SHADE TEMPERATURES

ine	0000	0200	0400	0600	0800	1000	1200	1400	1600	1800	2000	2200	Hrs.
in.	61.5	1	63.3 56 70	61.5 54 69	64.2 54 75	71.5 62 82	77.8 73 87	1	80.5 69.5 92	i	72.4 66 81	69.0 65 78	ole ole

Monthly Characteristics

Average 70.8°F Minimum 54. °F Maximum 92 °F (0800 hrs 2 Merch) (1600 hrs 28 March)

STATISTICAL RESULTS

FRERUARY 1947

Temperature at 2 Hourly intervals

ABADAM INCOMING CHUDE TELL'ERATURES

(SURFACE LINE SYSTEM)

Territor	0000	0200			0800	1000	1200	1400	1600	1800	2000	2200	Hrs.
7834		51.2	:		ļ	57.2	69.7	77.4	79.4	7406	65.8	59.5	Oğ.
in.	16	43	39	37	36	42	56	57	57	55	56	52	್ರೌ
100-	6 6	63	60	59	59	66	80	87	91	88	77	72	og

Diurnal Temperature Range

Monthly Characteristics

Average 35,60% Smisson 5 op Nextson 50 or Average 60.95 F
Minimum 36 F (0800 hrs 7 Feb.)
Haximum 91 F (1600 hrs 17 Feb.)

ABADAN SHADE TEMPERATURES

ime	9000	0200	04.00	0600	0 800	1000	1200	1400	1600	1800	2000		öre.
1 8 9-	56.5	5 4 .8	53.2	51.8	51.6	58.7	65 ₀ 3	67.8	68.7	66.1	61.8	58∘9	o _F
an.	49	48	46	ist.	4.5	50	56	58	58.5	57	55	51.5	Sp.
ex.	63	64	62	61.5	63	66	74	76	76	72	69	68	Op.

Monthly Characteristics

Average 59.60P

Minimum 43 °F (0800 hrs 7 Feb.)
Maximum 76 °F (1400 hrs 27 Feb.)

JANUARY 1947

STATISTICAL RESULTS

Temperature at 2 Hourly intervals

ARADAM INCOMING CRUDE TEMPERATURES

(SURFACE LINE SYSTEM)

								-				AND DESCRIPTIONS OF THE PERSONS ASSESSMENT ASSESSMENT OF THE PERSONS ASSESSMENT OF THE PERSONS ASSESSMENT OF THE PERSONS A	ì
3562	6000	0200	0400	0600	0800	1000	1200	1400	1600	1860	2000	2200	Erse
∂rge.,	55.5	51.1	49.2	48.1	48.1	5 5.6	6 5.8	72.5	72.6	67.,5	61.0	56.6	op
.20	işi.	i,2	40	39	40	47	56	60	60	59	50	46	0 3)
. II.	55	65	63.	60	60	69	78	82	84	80	70	64.	Op

Lawrent Temperature Range

Monthly Characteristics

Average 25.5°B Average 5 op Maximum 43 op

Average 58.5°F
Mindmum 59 °F (0600 hrs 27 Jan.)
Maximum 84 °F (1600 hrs 12 Jan.)

ABADAM SHADE TEMPERATURES

- me	0000	0200	0400		0800	1000	1300	1400		1800	2000	2200	W.s.
γge	55.9	54.7	53.2	52.1	52 , 2	58.4	63.5	65.6	6 6.0	62.9	59.7	57 _° 5	O _j a
n.	50	58	1,6	46	46	52.	56	57	60	59	52	51	C.J.
-UE.	65	63	64 <u>.</u>	64	67	70	72	73	73	69	67	66	OFF

Monthly Characteristics

Average 58.4°F Minimum 46 °F (0400 hrs 18 Jan.) Meximum 73 °F (1400 hrs 31 Jan.)

DECEMBER 1946

STATISTICAL RESULES

Temperature at 2 hourly intervals

ABADAN INCOMING CRUITE TEMPERATURES

(SUMPACE LINE STATEM)

	To the substitute of the subst		1	0600	0600	1000		1400	1600	1800	5000	2200	Exec.
3 . •	54.5	!		1		56,2	1			68.7	62.6	58.2	o _M
Æ	32	39	35	14	35	43	55	63	65	60	52	47	οħ
7.	65	60	60	59	59	65	78	84	87	80	78	70	O.M.

Enward Temperature Range

Average 24, 10F

Minimpa Maximum 39 of Monthly Characteristics

Average 59.50p Minimum 33 °F (0800 hrs 24 Dec.) Maximum 87 °F (1600 hrs 1 Dec.)